# Dispiroketals in synthesis. Part 25. ${ }^{1}$ Further reactions of dispiroketal protected glycolate to afford optically active 1,2,3,4-tetraols 

Morifumi Fujita, Dramane Lainé and Steven V. Ley *<br>Department of Chemistry, University of Cambridge, Lensfield Road, Cambridge, UK CB2 IEW

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#### Abstract

Glycolic acid can be converted to optically active 1,2,3,4-tetraols using a dispiroketal unit as a protecting group and chiral auxiliary. Aldol reactions of dispiroketal protected glycolate with aldehydes afford one diastereoisomer preferentially with two newly formed stereogenic centres. To extend the polyol chain, the carbonyl group of the aldol product is converted to a vinyl ether by the Tebbe reagent after protection of the free alcohol. A subsequent hydroboration-oxidation protocol affords the dispiroketal protected tetraol. The final deprotection of the tetraol occurs selectively without epimerisation or migration of the silyloxy protecting groups.


## Introduction

Sequences of 1,2-diols are observed in many natural products of biological significance, ranging from simple monosaccharides $^{2}$ to palytoxin. ${ }^{3}$ The development of methods for the stereocontrolled preparation of polyols is therefore essential for the synthesis of polyhydroxylated natural products. ${ }^{4}$ At present, the selective protection and deprotection protocols of hydroxy groups of polyols are necessary for successful polyol synthesis.
In recent years, our group has developed the use of bi(dihydropyran) derivatives for the regio- and stereoselective protection of 1,2 -diols as their corresponding dispiroketals. ${ }^{5}$ Chiral bi(dihydropyran)s are versatile and offer the possibility of enantioselectively desymmetrising meso polyols, thermodynamically resolving racemic 1,2 -diols, and selectively protecting diequatorial vicinal diols in sugars. ${ }^{6}$ The dispiroketal protecting group in sugars may also play an important role for the reactivity tuning in glycoside coupling reactions. ${ }^{7}$ Dispiroketals are not only used as protecting groups, they may also be synthetically useful as chiral auxiliaries. ${ }^{8}$

Here, we report the enantiodifferentiating preparation of 1,2,3,4-tetraols from a dispiroketal protected glycolate. The dispiroketal unit acts both as a protecting group and a chiral auxiliary to yield a dispiroketal protected tetraol which is then selectively deprotected. Thus, the dual-function of dispiroketals provides an opportunity to assemble valuable protected homochiral polyols in an efficient fashion.

## Results and discussion

The dispiroketal protected glycolate $\mathbf{2}$ was obtained as a racemate in $67 \%$ yield by treating bi(dihydropyran) $\mathbf{1}$ and glycolic acid with a catalytic quantity of $\mathrm{Ph}_{3} \mathrm{P} \cdot \mathrm{HBr}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ for 17 h at room temperature (Scheme 1). The dispiroketal 2 was obtained as a single racemic diastereoisomer with maximum anomeric stabilisation. The protected glycolate was then deprotonated by treatment with LDA in THF at $-78^{\circ} \mathrm{C}$ and reacted in a highly diastereodifferentiating manner with benzaldehyde in the presence of 1,3-dimethyl-3,4,5,6-tetrahydro-2(1H)-pyrimidone (DMPU) to give only 2 diastereoisomers out of four possible diastereoisomers ${ }^{9}$ (Table 1). The structure of the major isomer $\mathbf{5 a}$, formed as a racemate in $89 \%$ yield, which has an equatorial alkyl group and is an erythro diol, was determined by X-ray


Scheme 1 Reagents and conditions: i, $\mathrm{Ph}_{3} \mathrm{P} \cdot \mathrm{HBr}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{rt}, 17 \mathrm{~h}, 67 \%$; ii, glycolic acid, $\mathrm{Ph}_{3} \mathrm{P} \cdot \mathrm{HBr}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, rt, $6 \mathrm{~h}, 94 \%$.
crystallography (Fig. 1). $\dagger$ The minor diastereoisomer 5b ( $2 \%$ yield), which has an axial alkyl group, was assigned from evidence of a strong NOE between the axial proton on C-2 and the benzyl proton but the configuration of the newly formed hydroxy group remained, however, undetermined. The large preference for one diastereomer could be rationalised by considering the chair-like six-membered transition state ${ }^{10}$ whereby the alkyl group of the aldehyde was directed pseudoequatorially in the transition state (Scheme 2). Extension of the reaction time to 3 h decreased the product yield to only $58 \%$. Interestingly, when the aldol reaction was quenched after warming up to room temperature, dehydration occurred to give the $Z$ olefin 9 as the only product in $19 \%$ yield (Scheme 3). The stereochemistry of $\mathbf{9}$ was determined by NOESY ${ }^{1} \mathrm{H}$ NMR. This result clearly indicated that the aldol reaction needed to be worked up at $-78^{\circ} \mathrm{C}$ to minimise the syn-dehydration and potential retro-aldol products.
The aldol product 5 a could be deprotected by treatment with camphorsulfonic acid and ethylene glycol in methanol to give

[^0]Table 1 Stereodifferentiating aldol reaction of dispiroketal protected glycolate

|  |  |  |
| :---: | :---: | :---: |
|  |  |  |
| $2 \mathrm{X}=\mathrm{H}$ | $\begin{aligned} & 5 \mathrm{X}=\mathrm{H}, \mathrm{R}=\mathrm{Ph} \\ & 6 \mathrm{X}=\mathrm{H}, \mathrm{R}=\mathrm{CH}_{2}=\mathrm{CH} \end{aligned}$ |  |
| $4 \mathrm{X}=\mathrm{CH}_{2} \mathrm{SPh}$ | $7 \mathrm{X}=\mathrm{CH}_{2} \mathrm{SPh}, \mathrm{R}=\mathrm{Ph}$ |  |
| Dispiroketal | RCHO | Products (Yield, \% |
| 2 | $\mathrm{R}=\mathrm{Ph}$ | $\mathbf{5 a}(89), \mathbf{5 b}$ (2) |
| 2 | $\mathrm{R}=\mathrm{CH}_{2}=\mathrm{CH}$ | 6 a (78) |
| 4 | $\mathrm{R}=\mathrm{Ph}$ | 7 a (84) |
| 4 | $\mathrm{R}=\mathrm{CH}_{2}=\mathrm{CH}$ | 8a (78), 8b (3) |
| 4 | $\mathrm{R}=\mathrm{Bu}^{t}$ | No adduct |



Fig. 1 X-Ray structure of aldol product 5a.


Scheme 2 Transition states of the aldol reaction.
methyl 2,3-dihydroxy-3-phenylpropionoate $\mathbf{1 0}$ and ethylene glycol dispiroketal 12 in high yields (Table 2). The diol 10 was obtained as a single racemic diastereoisomer, whose relative

Table 2 Deprotection of the aldol products

5a $\mathrm{X}=\mathrm{H}, \mathrm{R}=\mathrm{Ph}$
$10 \mathrm{R}=\mathrm{Ph}$
$12 \mathrm{X}=\mathrm{H}$
6a $X=H, R=\mathrm{CH}_{2}=\mathrm{CH}$
$11 \mathrm{R}=\mathrm{CH}_{2}=\mathrm{CH}$
7a $X=\mathrm{CH}_{2} \mathrm{SPh}, \mathrm{R}=\mathrm{Ph}$
$13 \mathrm{X}=\mathrm{CH}_{2} \mathrm{SPh}$

8a $X=\mathrm{CH}_{2} \mathrm{SPh}, \mathrm{R}=\mathrm{CH}_{2}=\mathrm{CH}$

| Reactant | Products (yield, \%) |
| :---: | :--- |
| $\mathbf{5 a}^{a}$ | $\mathbf{1 0}(95)^{a} \mathbf{1 2}(94)$ |
| $\mathbf{6 a}{ }^{a}$ | $\mathbf{1 1}(62)^{a} \mathbf{1 2}(79)$ |
| $\mathbf{8 a}$ | $\mathbf{1 0}(99)^{b} \mathbf{1 3}(98)$ |
| $\mathbf{8 a}$ | $\mathbf{1 1}(81)^{b} \mathbf{1 3}(96)$ |
|  |  |
| ${ }^{a}$ Racemic mixture. ${ }^{b}$ Enantiopurity excess is greater than $95 \%$. |  |



Scheme 3 Reagents and conditions: i, $\operatorname{Pr}_{2}{ }_{2} \mathrm{NH}, \mathrm{Bu}{ }^{\mathrm{n}} \mathrm{Li}, \mathrm{THF}, \mathrm{DMPU}$, then PhCHO at $-78^{\circ} \mathrm{C}$ to rt, overnight, $19 \%$.
stereochemistry was determined to be erythro by comparison with the spectroscopic properties to those reported in the literature. ${ }^{11-13}$ Importantly, these results indicated that the deprotection step proceeds without epimerisation. When acrolein was used instead of benzaldehyde, the stereoselective aldol product $6 \mathbf{a}$ was also obtained and deprotected readily using similar procedures to give the racemic anti diol 11.

The above results suggested that enantiodifferentiating aldol reactions could be achieved potentially using chiral dispiroketals. Optically active dispiroketal 4 was prepared from enantiomerically pure bi(dihydropyran) 3 (Scheme 1) as a single diastereoisomer with the side chain phenylthiomethyl substituents equatorial, as indicated from the NOE signal observed between $15-\mathrm{H}$ and $2-\mathrm{H}$. The aldol reaction of the enolate derived from 4 gave preferentially one diastereoisomer using benzaldehyde, acrolein and acetaldehyde, respectively. However, in the case of pivalaldehyde, no adduct was obtained and only the starting material $\mathbf{4}$ was recovered. It was subsequently found that the reaction occurs when HMPA was used instead of DMPU as a co-solvent. ${ }^{14}$ The deprotection of the benzaldehyde reaction product 7 a in methanol in the presence of camphorsulfonic acid (CSA) and ethylene glycol gave methyl $(+)$-erythro-2,3-dihydroxy-3-phenylpropionoate $\mathbf{1 0}$. After comparison of the optical rotation with literature values, ${ }^{13}$ the absolute stereochemistry of $\mathbf{1 0}$ was assigned as $(2 S, 3 S)$. The absolute structure of the deprotected diol $\mathbf{1 0}$ indicated that the aldol product 7a had the same relative structure as racemic aldol product 5a, whose relative stereochemistry was determined by X-ray crystallography. In agreement with the above evidence, the ee of $\mathbf{1 0}$ was determined to be greater than $95 \%$ (i.e. essentially enantiopure) by analysis of the corresponding Mosher ester. Methyl ( $2 S, 3 S$ )-2,3-dihydroxypent-4-enoate $\mathbf{1 1}$
was also obtained in enantiomerically pure form (ee $>95 \%$ ) from 8a.

2,3-O-Isopropylidene-d-glyceraldehyde, which is readily accessible from inexpensive D-manitol, has been widely used as an important chiral $C_{3}$ building block. ${ }^{15}$ By use of this homochiral aldehyde in an aldol reaction with a chiral dispiroketal unit, we allowed the effect of double asymmetric induction ${ }^{16,17}$ on the stereoselectivity to be examined. The aldol reaction of the D-glyceraldehyde derivative $\mathbf{1 4}$ with the dispiroketal protected glycolate $\mathbf{4}$ gave products in low yield and poor stereoselectivity, as depicted in Scheme 4. The major stereoisomer 15,




18


19
20
Scheme 4 Reagents and conditions: i, $\operatorname{Pr}^{i}{ }_{2} \mathrm{NH}, \mathrm{Bu}^{\mathrm{n}} \mathrm{Li}$, THF, DMPU, then 14 at $-78{ }^{\circ} \mathrm{C}, 30 \mathrm{~min}$, $14 \%$ ( $\mathbf{1 5 : 1 6}$ 6:4, mixture); ii, glycolic acid, $\mathrm{Ph}_{3} \mathrm{P} \cdot \mathrm{HBr}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, rt, $8 \mathrm{~h}, 86 \%$; iii, $\operatorname{Pr}_{2}^{\mathrm{i}} \mathrm{NH}, \mathrm{Bu}^{\mathrm{n}} \mathrm{Li}$, THF, DMPU, then 14 at $-78{ }^{\circ} \mathrm{C}, 60 \mathrm{~min}(19: 54 \%, \mathbf{2 0}: 9 \%$, the other 2 diastereomers mixture: $4 \%$, recovered 18: $14 \%$ ).
expected from a chair-like six-membered transition state, had an unfavourable stereostructure as predicted by Felkin's nonchelation model ${ }^{18}$ on nucleophilic additions of D-glyceraldehyde (Scheme 5). ${ }^{19}$ The mismatched pair of the chiral enolate and the chiral aldehyde could be avoided by using the ( + ) form dispiroketal protected glycolate 18 prepared from ( $S, S$ ) $-2,2^{\prime}$ bis(phenylthiomethyl)bi(dihydropyran) 17. As expected, the aldol reaction using 18 afforded 19 , which had an equatorial alkyl group as judged from the strong NOE spectrum between $2-\mathrm{H}$ and $15-\mathrm{H}$, as the major isomer in $54 \%$ yield (Scheme 4). Minor isomers, however, were also obtained in significant quantities. The minor isomer $\mathbf{2 0}$ also had an equatorial alkyl group as judged from the NOE signal between $2-\mathrm{H}$ and $15-\mathrm{H}$. The formation of the unpredicted stereoisomer 20 by a chair-like six-membered transition state, suggested the contribution of





Major


Scheme 5 Stereoselectivity of an aldol reaction of 2,3-O-isopropyl-idene-d-glyceraldehyde with an enolate (ref. 18).
another chelation model ${ }^{20}$ for the aldol transition state. The co-ordination of a lithium ion to an $\alpha$ - or $\beta$-oxygen of glyceraldehyde may threaten the preference of the chair-like sixmembered transition state.

The aldol reactions mentioned above afforded the stereocontrolled diols. Further reactions from these aldol products were performed in order to extend the polyol chain in a stereocontrolled manner. Thus, the carbonyl group of the racemic aldol product 5 a was converted to the vinyl ether $\mathbf{2 3}$ by using the Tebbe reagent after protection of the free alcohol moiety of 5a. A subsequent two-step hydroboration-oxidation sequence from $\mathbf{2 3}$ afforded the alcohols $\mathbf{2 5}$ and $\mathbf{2 6}$ as a mixture of isomers in which $\mathbf{2 5}$ is the major product (Scheme 6). These compounds were easily separated by chromatography and their stereochemistry was readily determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy. The axial alcohol 25 was formed preferentially by hydroboration from the least hindered olefin face. ${ }^{21}$
Similar procedures were applied to the enantiomerically pure aldol product 7 a to yield $\mathbf{2 7}$ as an optically active compound (Scheme 6). Furthermore, oxidation of the phenylthiomethyl groups at the 2- and 9-positions lead to selective deprotection of the dispiroketal moiety via base promoted $\beta$-elimination of bis-sulfone 29 generated by oxidation. ${ }^{5 c}$ Deprotection under mild conditions afforded the optically active tertraol $\mathbf{3 0}$ without migration of the TBS group.

In summary we have developed a preparative method for the enantiodifferentiating synthesis of 1,2,3,4-tetraols via 2,3dihydroxycarboxylates using the versatile dispiroketal unit as a protecting group and a chiral auxiliary. The dispiroketal protected polyols can be deprotected selectively without epimerisation and migration of silyloxy protecting groups.

## Experimental

Proton NMR spectra were recorded on a Bruker DRX-600 ( 600 MHz ) or DPX-200 ( 200 MHz ) spectrometer as solutions in $\mathrm{CDCl}_{3}$ using the residual $\mathrm{CHCl}_{3}$ as an internal reference ( 7.26 ppm ). Coupling constants $J$ are quoted in $\mathrm{Hz} .{ }^{13} \mathrm{C}$ NMR spectra were recorded on a Bruker DRX-600 ( 150 MHz ), AM-400 ( 100 MHz ) or DPX-200 $(50 \mathrm{MHz})$ spectrometer as solutions in $\mathrm{CDCl}_{3}$. ${ }^{19} \mathrm{~F}$ NMR spectra were recorded on a Bruker AC-250 $(150 \mathrm{MHz})$ spectrometer as solutions in $\mathrm{CDCl}_{3}$. Infra-red spectra were recorded as thin films between sodium chloride plates deposited from $\mathrm{CDCl}_{3}$ solution on a Perkin-





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Scheme 6 Reagents and conditions: i, TBSCl, imidazole, DMF, rt, 21: 98\%, 22: 67\%; ii, Tebbe reagent, THF, pyridine, rt, 23: 96\%, 24: 88\%; iii, 9-BBN, THF, rt, 3 h , then $\mathrm{NaOH}, \mathrm{H}_{2} \mathrm{O}_{2}$, rt, 30 min , from 23 to yield 25 $(86 \%)$ and $26(14 \%)$, from 24 to yield $27(60 \%)$ and $28(20 \%)$; iv, MCPBA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, rt, $1.5 \mathrm{~h}, 88 \%$; v, LiN(TMS) ${ }_{2}$, THF, rt, $30 \mathrm{~min}, 53 \%$.

Elmer 1600 series FTIR spectrometer. Mass spectra were recorded at the EPSRC mass spectrometry service, Swansea. Microanalyses were performed in the University of Cambridge microanalysis laboratory. X-Ray crystallographic data were obtained by Cambridge University Department of Chemistry crystallographic service. Melting points were determined on a Reicher hot stage apparatus and are uncorrected. Optical rotations were measured using an Optical Activity AA-1000 polarimeter and are quoted in $10^{-1} \mathrm{deg} \mathrm{cm}^{2} \mathrm{~g}^{-1}$. Flash column chromatography was performed on Merck 9385 Kieselgel 60 silica (230-400 mesh) using hexane, 40-60 petroleum ether (petrol), diethyl ether (ether) or ethyl acetate as eluents. The reactions were monitored by thin layer chromatography and the plates were visualised with UV light ( 254 nm ) and acidified ammonium molybdate(IV).

All reactions were carried out under an argon atmosphere with dry freshly distilled solvents, under anhydrous conditions. Tetrahydrofuran (THF), diethyl ether (ether) and pentane were distilled from sodium-benzophenone and dichloromethane (DCM), triethylamine ( $\mathrm{NEt}_{3}$ ), benzene and toluene were distilled from calcium hydride. Dimethylformamide (DMF) was dried over 4 Å molecular sieves.

Preparation of 2,2'-bis(phenylthiomethyl)-3,3',4,4'-tetrahydro-6,6'-bi-2H-pyran 3 and 17
Chiral 2,2'-bis(phenylthiomethyl)-3, $3^{\prime}, 4,4^{\prime}$-tetrahydro-6,6'-bi$2 H$-pyran 3 and 17 were prepared according to literature procedures. ${ }^{6 a}$ By using $99.99 \% \mathrm{Pd}(\mathrm{MeCN})_{2} \mathrm{Cl}_{2}$ (Aldrich) and $99.995 \%$ anhydrous $\mathrm{CuCl}_{2}$ (Aldrich), the coupling reaction of 2-triisopropylsilyloxymethyl-3,4-dihydro-2H-pyran proceeded in $53 \%$ yield. The formed $2,2^{\prime}$-bis(triisopropylsilyloxymethyl)$3,3^{\prime}, 4,4^{\prime}$-tetrahydro-6, $6^{\prime}$-bi- $2 H$-pyran was enriched in optical purity by recrystallisation from ethanol. The conversion of triisopropylsilyloxymethyl groups to phenylthiomethyl groups was performed using the literature method. ${ }^{6 a}$

## Preparation of 2,3- $O$-isopropylidene-D-glyceraldehyde 14

The title compound was prepared from $\mathrm{C}-\mathrm{C}$ bond cleavage of $1,2: 5,6$-di- $O$-isopropylidene-D-mannitol by $\mathrm{Pb}(\mathrm{OAc})_{4}$ in benzene. ${ }^{22}$ The formed aldehyde was unstable due to the tendency for polymerisation. Thus, the aldehyde was used for aldol reactions immediately after distillation under reduced pressure.
$\left(6 R^{*}, 7 R^{*}\right)-1,8,13,16-T e t r a o x a d i s p i r o[5.0 .5 .4]$ hexadecan-14-one 2

To a solution of bi(dihydropyran) $\mathbf{1}(1.0 \mathrm{~g}, 6.0 \mathrm{mmol})$ and glycolic acid ( $0.68 \mathrm{~g}, 8.9 \mathrm{mmol}, 1.5$ eq.) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(50 \mathrm{ml})$ was added $\mathrm{Ph}_{3} \mathrm{P} \cdot \mathrm{HBr}(0.2 \mathrm{~g}, 0.6 \mathrm{mmol}, 0.1 \mathrm{eq}$.$) . The mixture was$ stirred at room temperature for 17 h during which time the mixture became yellow-orange, then the solvent was evaporated in vacuo. Purification of the residue by flash chromatography (eluent: ether-petrol $1: 3$ ) gave the title compound 2 $(0.98 \mathrm{~g}, 4.0 \mathrm{mmol}, 67 \%)$ as a white solid; $\mathrm{mp} 103{ }^{\circ} \mathrm{C}$ (etherpetrol) (Found: C, 59.47; H, 7.48. $\mathrm{C}_{12} \mathrm{H}_{18} \mathrm{O}_{5}$ requires C, 59.49; H, 7.49\%); $v_{\max }($ film $) / \mathrm{cm}^{-1} 2954,1748,1602,1522,1441,1358$, $1295,1255,1072,984 ; \delta_{\mathrm{H}}\left(200 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 4.33(1 \mathrm{H}, \mathrm{d}, J 17.6$, $15-\mathrm{H}), 4.25(1 \mathrm{H}, \mathrm{d}, J 17.6,15-\mathrm{H}), 4.06-3.46(4 \mathrm{H}, \mathrm{m}, 2-\mathrm{H} \times 2,9-$ $\mathrm{H} \times 2), 2.03-1.54(12 \mathrm{H}, \mathrm{m}) ; \delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 167.3(\mathrm{C}=\mathrm{O})$, 103.3, 95.0 (6-C, 7-C), 62.9, 62.3 (2-C, 9-C), 59.5 (15-C), 28.4, $27.8,24.5,24.3,17.9,17.1$ (3-C, 4-C, 5-C, 10-C, 11-C, 12-C); $m / z(\mathrm{CI}) 260\left(\mathrm{MNH}_{4}{ }^{+}\right), 243\left(\mathrm{MH}^{+}, 11 \%\right), 167(100)$ [Found $\left(\mathrm{MH}^{+}\right)$243.1232. $\mathrm{C}_{12} \mathrm{H}_{19} \mathrm{O}_{5}$ requires $M \mathrm{H}, 243.1232$ ].

## General procedure for aldol reactions

A solution of LDA in THF ( 2.5 ml ) was prepared under argon from diisopropylamine ( 1.1 eq.) and $\mathrm{Bu}^{\mathrm{n}} \mathrm{Li}(1.6 \mathrm{M}$ solution in hexane, 1.1 eq.). After stirring for 20 min and cooling to $-78^{\circ} \mathrm{C}$, a solution of the dispiroketal protected glycolate in THF ( 1.0 ml ) was added via cannula, the flask being rinsed with THF ( $2 \times 0.75 \mathrm{ml}$ ). After stirring for 30 min at $-78^{\circ} \mathrm{C}$, a second portion of $\mathrm{Bu}^{\mathrm{n}} \mathrm{Li}$ solution (1.1 eq.) and 1,3-dimethyl-3,4,5,6-tetrahydro-2(1H)-pyrimidone (DMPU) $(0.5 \mathrm{ml})$ as a co-solvent were added. The mixture was allowed to stir for 30 $\min$ at $-78^{\circ} \mathrm{C}$, then the appropriate aldehyde ( 2.0 eq.) was introduced. The reaction was quenched at $-78^{\circ} \mathrm{C}$ by the addition of saturated aqueous ammonium chloride after stirring for 30 min . The mixture was then allowed to warm up to room temperature and water added to dissolve the precipitated salts. The phases were separated and the aqueous layer was extracted with ether $(\times 3)$. The combined organic extracts were dried $\left(\mathrm{MgSO}_{4}\right)$, evaporated in vacuo and the residue purified by flash chromatography (eluent: ether-petrol $1: 3 \rightarrow 1: 1$; gradient) to give the desired compound.
( $\left.6 R^{*}, 7 R^{*}, 15 S^{*}\right)-15-\left[\left(S^{*}\right)\right.$-(Hydroxy)phenylmethyl $]-1,8,13,16-$ tetraoxadispiro[5.0.5.4]hexadecan-14-one 5 a and $\left(6 R^{*}, 7 R^{*}\right.$, $\left.15 R^{*}\right)-15-\left[\left(R^{*}\right)\right.$-(hydroxy)phenylmethyl]-1,8,13,16-tetraoxadi-spiro[5.0.5.4]hexadecan-14-one 5b. Following the general procedure with the dispiroketal protected glycolate $2(120 \mathrm{mg}, 0.50$ $\mathrm{mmol})$ and benzaldehyde $(0.1 \mathrm{ml}, 1.0 \mathrm{mmol})$, a mixture of 5 a and $\mathbf{5 b}$ ( $158 \mathrm{mg}, 0.454 \mathrm{mmol}, 91 \%$ ) was obtained after flash chromatography (eluent: ether-petrol $1: 2$ ). The ratio of 5a and
$\mathbf{5 b}$ was determined by ${ }^{1} \mathrm{H}$ NMR as $97.5: 2.5$, which gives a de of $95 \%$. Careful flash chromatography (eluent: ether-petrol $1: 3 \rightarrow 1: 1$; gradient) allows the separation of the mixture of $\mathbf{5 b}$ and $\mathbf{5 a}$ (in order of elution).
5a: mp 117-119 ${ }^{\circ} \mathrm{C}$ (ether) (Found: C, 65.56; H, 6.91. $\mathrm{C}_{19} \mathrm{H}_{24}{ }^{-}$ $\mathrm{O}_{6}$ requires $\mathrm{C}, 65.50 ; \mathrm{H}, 6.94 \%$ ); $v_{\text {max }}($ film $) / \mathrm{cm}^{-1} 3460,2953$, $1744,1602,1454,1356,1072 ; \delta_{\mathrm{H}}\left(600 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.43(2 \mathrm{H}$, d, $J 7.5, \mathrm{Ar}), 7.34(2 \mathrm{H}, \mathrm{t}, J 7.5, \mathrm{Ar}), 7.28(1 \mathrm{H}, \mathrm{t}, J 7.5, \mathrm{Ar}), 5.13$ $(1 \mathrm{H}, \mathrm{d}, J 3.7, \mathrm{PhC} H), 4.35(1 \mathrm{H}, \mathrm{d}, J 5.1,15-\mathrm{H}), 4.04(1 \mathrm{H}, \mathrm{d}$, $J 1.9, \mathrm{OH}), 3.92-3.87\left(1 \mathrm{H}, \mathrm{m}, 9_{\mathrm{eq}}-\mathrm{H}\right), 3.74-3.70\left(2 \mathrm{H}, \mathrm{m}, 9_{\mathrm{ax}}-\mathrm{H}\right.$, $\left.2_{\mathrm{eq}}-\mathrm{H}\right), 3.48\left(1 \mathrm{H}, \mathrm{td}, J 11.4,3.2,2_{\mathrm{ax}}-\mathrm{H}\right), 1.98-1.46(12 \mathrm{H}, \mathrm{m})$; $\delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 167.4(\mathrm{C}=\mathrm{O}), 138.5,127.8,127.8,126.8$ (Ar), 103.5, 95.6 (6-C, 7-C), 74.3, 73.9 (15-C, PhCH), 62.8, 62.4 (2-C, 9-C), 28.4, 27.7, 24.4, 24.2, 17.8, 17.1 (3-C, 4-C, 5-C, 10-C, 11-C, 12-C); m/z (CI) $366\left(\mathrm{MNH}_{4}^{+}, 6 \%\right), 349\left(\mathrm{MH}^{+}, 3\right), 243$ (13), 167 (100) [Found $\left(\mathrm{MH}^{+}\right) 349.1651 . \mathrm{C}_{19} \mathrm{H}_{25} \mathrm{O}_{6}$ requires $M \mathrm{H}, 349.1651$ ].
5b: $v_{\text {max }}($ film $) / \mathrm{cm}^{-1} 3473,2949,2360,1722,1454,1440,1354$, 1288, 1240, 1214, 1184, 1108, 1074, 1052, 989, 964, 912, 760, $731,700,668 ; \delta_{\mathrm{H}}\left(600 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.48(2 \mathrm{H}, \mathrm{d}, J 7.4, \mathrm{Ar}), 7.37$ ( $2 \mathrm{H}, \mathrm{t}, J 7.4, \mathrm{Ar}), 7.32(1 \mathrm{H}, \mathrm{t}, J 7.3, \mathrm{Ar}), 4.93(1 \mathrm{H}, \mathrm{d}, J 8.7$, $\mathrm{PhCH}), 4.49(1 \mathrm{H}, \mathrm{br}, \mathrm{OH}), 4.40(1 \mathrm{H}, \mathrm{d}, J 8.7,15-\mathrm{H}), 3.93-3.83$ $(3 \mathrm{H}, \mathrm{m}, 9-\mathrm{H}, 2-\mathrm{H}), 3.76-3.73(1 \mathrm{H}, \mathrm{m}, 9-\mathrm{H}$ or $2-\mathrm{H}), 2.03-1.91$ $(2 \mathrm{H}, \mathrm{m}), 1.75-1.26(10 \mathrm{H}, \mathrm{m}) ; \delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 172.1(\mathrm{C}=\mathrm{O})$, 139.3, 128.1, 128.0, 127.4 (Ar), 104.4, 96.2 (6-C, 7-C), 74.2, 72.5 (15-C, PhCH), 63.1, 62.1 (2-C, 9-C), 28.5, 28.4, 24.8, 24.2, 17.7, 17.4 (3-C, 4-C, 5-C, 10-C, 11-C, 12-C); $m / z$ (CI) 366 ( $\mathrm{MNH}_{4}{ }^{+}$, $2 \%), 349\left(\mathrm{MH}^{+}, 6\right), 243$ (12), 167 (100) [Found ( $\mathrm{MH}^{+}$) 349.1651. $\mathrm{C}_{19} \mathrm{H}_{25} \mathrm{O}_{6}$ requires $M \mathrm{H}, 349.1651$ ].
( $6 R^{*}, 7 R^{*}$ )-( $Z$ )-15-Benzylidene-1,8,13,16-tetraoxadispiro-[5.0.5.4]hexadecan-14-one 9. The same general procedure for aldol reactions ( $2: 122 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) was applied except that the reaction was quenched after warming up to room temperature to give the title compound $9(31.5 \mathrm{mg}, 19 \%)$; $v_{\text {max }}($ film $) /$ $\mathrm{cm}^{-1} 3431,2951,1725,1631,1376,1363,1270,1182,1075,995 ;$ $\delta_{\mathrm{H}}\left(600 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.82(2 \mathrm{H}, \mathrm{d}, J 7.5, \mathrm{Ar}), 7.38(2 \mathrm{H}, \mathrm{t}, J 7.5$, Ar), 7.31 ( $1 \mathrm{H}, \mathrm{t}, J 7.5, \mathrm{Ar}$ ), $7.00(1 \mathrm{H}, \mathrm{s}, \mathrm{PhCH}=), 4.02-3.97(1 \mathrm{H}$, $\mathrm{m}, 9-\mathrm{H}$ or $2-\mathrm{H}), 3.86-3.82(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}$ or $9-\mathrm{H}), 3.73-3.70(1 \mathrm{H}$, $\mathrm{m}, 9-\mathrm{H}$ or $2-\mathrm{H}), 3.62\left(1 \mathrm{H}, \mathrm{td}, J 11.6,3.6,2_{\mathrm{ax}}-\mathrm{H}\right.$ or $\left.9_{\mathrm{ax}}-\mathrm{H}\right), 2.14-$ $2.00(3 \mathrm{H}, \mathrm{m}), 1.89-1.84(1 \mathrm{H}, \mathrm{m}), 1.77-1.57(8 \mathrm{H}, \mathrm{m}) ; \delta_{\mathrm{C}}(100$ $\left.\mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 160.5(\mathrm{C}=\mathrm{O}), 137.0,133.5,128.6,128.5$ ( Ar ), $118.9(\mathrm{PhCH}=), 102.3,97.5(6-\mathrm{C}, 7-\mathrm{C}), 63.0,62.5(2-\mathrm{C}, 9-\mathrm{C})$, 28.5, 27.5, 24.5, 24.4, 18.5, 17.3 (3-C, 4-C, 5-C, 10-C, 11-C, $12-\mathrm{C}) ; \mathrm{m} / \mathrm{z}(\mathrm{CI}) 348\left(\mathrm{MNH}_{4}^{+}, 62 \%\right), 331\left(\mathrm{MH}^{+}, 98\right), 167(100)$ [Found $\left(\mathrm{MH}^{+}\right) 331.1545 . \mathrm{C}_{19} \mathrm{H}_{23} \mathrm{O}_{5}$ requires $M \mathrm{H}, 331.1545$ ].

## ( $1^{\prime} S^{*}, 6 R^{*}, 7 R^{*}, 15 S^{*}$ )-15-(1-Hydroxyprop-2-enyl)-1,8,13,16-

 tetraoxadispiro[5.0.5.4]hexadecan-14-one 6a. By the general procedure using $2(120 \mathrm{mg}, 0.5 \mathrm{mmol})$ and acrolein ( $74 \mathrm{ml}, 1.0$ $\mathrm{mmol})$, the title compound $\mathbf{6 a}(115 \mathrm{mg}, 78 \%)$ was obtained as a white solid after flash chromatography (eluent: ether-petrol $1: 3 \rightarrow 1: 1$; gradient); $\mathrm{mp} 88^{\circ} \mathrm{C}$ (ether-petrol) (Found: $\mathrm{C}, 60.63$; $\mathrm{H}, 7.45 . \mathrm{C}_{15} \mathrm{H}_{22} \mathrm{O}_{6}$ requires $\left.\mathrm{C}, 60.39 ; \mathrm{H}, 7.43 \%\right)$; $v_{\max }($ film $) / \mathrm{cm}^{-1}$ 3480, 2952, 2877, 1746, 1641, 1570, 1440, 1375, 1355, 1290, 1251, 1215, 1186, 1107, 1072, 995, 965, 884, 735; $\delta_{\mathrm{H}}(600 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 6.02\left(1 \mathrm{H}\right.$, ddd, $\left.J 17.2,10.5,6.4,2^{\prime}-\mathrm{H}\right), 5.40(1 \mathrm{H}, \mathrm{d}$, $\left.J 17.2,3^{\prime}-\mathrm{H}\right), 5.26\left(1 \mathrm{H}, \mathrm{d}, J 10.5,3^{\prime}-\mathrm{H}\right), 4.52\left(1 \mathrm{H}, \mathrm{br}, 1^{\prime}-\mathrm{H}\right)$, $4.25(1 \mathrm{H}, \mathrm{d}, J 3.8,15-\mathrm{H}), 4.00-3.95\left(1 \mathrm{H}, \mathrm{m}, 9_{\text {eq }}-\mathrm{H}\right), 3.82-3.76$ $\left(2 \mathrm{H}, \mathrm{m}, 9_{\mathrm{ax}}-\mathrm{H}, 2_{\mathrm{eq}}-\mathrm{H}\right), 3.67-3.62\left(1 \mathrm{H}, \mathrm{m}, 2_{\mathrm{ax}}-\mathrm{H}\right), 3.50(1 \mathrm{H}, \mathrm{br}$, $\mathrm{OH}), 2.00-1.79(4 \mathrm{H}, \mathrm{m}), 1.72-1.55(8 \mathrm{H}, \mathrm{m})$; $\delta_{\mathrm{C}}(50 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 166.9(\mathrm{C}=\mathrm{O}), 134.8\left(2^{\prime}-\mathrm{C}\right), 117.3\left(3^{\prime}-\mathrm{C}\right), 103.2,95.6$ (6-C, 7-C), 73.9, 73.7 ( $\left.15-\mathrm{C}, 1^{\prime}-\mathrm{C}\right), 62.9,62.5$ (2-C, 9-C), 28.5, 27.6, 24.5, 24.2, 17.9, 17.0 (3-C, 4-C, 5-C, 10-C, 11-C, 12-C); $\mathrm{m} / \mathrm{z}(\mathrm{CI}) 299\left(\mathrm{MH}^{+}, 2 \%\right), 167(100)$ [Found $\left(\mathrm{MH}^{+}\right) 299.1494$. $\mathrm{C}_{15} \mathrm{H}_{23} \mathrm{O}_{6}$ requires $M \mathrm{H}, 299.1495$ ].
## Methyl (2RS,3RS)-2,3-dihydroxy-3-phenylpropionoate 10

To a solution of $5 \mathbf{5}(98.3 \mathrm{mg}, 0.282 \mathrm{mmol})$ in anhydrous meth-
anol ( 5 ml ) was added CSA ( $18.8 \mathrm{mg}, 0.081 \mathrm{mmol}, 0.3$ eq.) and ethylene glycol ( $36 \mathrm{mg}, 0.58 \mathrm{mmol}, 2.0$ eq.). The mixture was heated under reflux for 4 h and after cooling to room temperature the solvent was evaporated in vacuo. Purification of the residue by flash chromatography (eluent: ether-petrol $1: 3 \rightarrow$ $1: 0$; gradient) gave in order of elution, the ethylene glycol dispiroketal 12 ( $60.3 \mathrm{mg}, 0.267 \mathrm{mmol}, 94 \%$ ), which had identical spectroscopic properties to those reported in literature, ${ }^{5}$ and the diol $\mathbf{1 0}(52.3 \mathrm{mg}, 0.260 \mathrm{mmol}, 95 \%)$. The melting point of $\mathbf{1 0}$ $\left(87-90^{\circ} \mathrm{C}\right)$ was in agreement with the one reported for the $(2 R S, 3 R S)$ form $\left(87^{\circ} \mathrm{C}\right)$, but not with the one of the $(2 R S, 3 S R)$ form $\left(69^{\circ} \mathrm{C}\right) .{ }^{10} \mathrm{H}$ NMR spectral data ${ }^{11,12}$ are also in agreement with that of the ( $2 R S, 3 R S$ )-diol. 10: $\mathrm{mp} 87-90^{\circ} \mathrm{C}$ (ether-petrol) (Found: C, $61.15 ; \mathrm{H}, 6.20 . \mathrm{C}_{10} \mathrm{H}_{12} \mathrm{O}_{4}$ requires C, 61.22; H, $6.16 \%) ; v_{\max }($ film $) / \mathrm{cm}^{-1} 3379,2930,1735,1602,1454,1384$, 1236, 1106, 1052; $\delta_{\mathrm{H}}\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.37-7.30(5 \mathrm{H}, \mathrm{m}, \mathrm{Ar})$, $5.02(1 \mathrm{H}, \mathrm{t}, J 5.1,3-\mathrm{H}), 4.51(1 \mathrm{H}, \mathrm{t}, J 5.2,2-\mathrm{H}), 3.70(3 \mathrm{H}, \mathrm{s}$, $\mathrm{Me}), 2.87(1 \mathrm{H}, \mathrm{d}, J 6.6, \mathrm{OH}), 2.84(1 \mathrm{H}, \mathrm{d}, J 6.1, \mathrm{OH}) ; \delta_{\mathrm{C}}(50$ $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 172.3 (C=O), 138.5, 128.3, 128.2, 126.3 (Ar), 75.0 (CH), $74.8(\mathrm{CH}), 52.4(\mathrm{Me}) ; m / z(\mathrm{CI}) 214\left(\mathrm{MNH}_{4}{ }^{+}, 100 \%\right), 196$ $\left(\mathrm{M}^{+}, 16 \%\right), 108$ (16) [Found $\left(\mathrm{MNH}_{4}{ }^{+}\right)$214.1079. $\mathrm{C}_{10} \mathrm{H}_{16} \mathrm{O}_{4} \mathrm{~N}$ requires $\left.M \mathrm{NH}_{4}, 214.1079\right]$.

## Methyl (2RS,3RS)-2,3-dihydroxypent-4-enoate 11

To a solution of $\mathbf{6 a}(69.3 \mathrm{mg}, 0.232 \mathrm{mmol})$ in anhydrous methanol ( 5 ml ) was added CSA ( $15.6 \mathrm{mg}, 0.067 \mathrm{mmol}, 0.3$ eq.) and ethylene glycol ( $34 \mathrm{mg}, 0.55 \mathrm{mmol}, 2.0$ eq.). The mixture was heated under reflux for 2 h and after cooling the solvent was evaporated in vacuo. Purification of the residue by flash chromatography (eluent: ether-petrol $1: 3 \rightarrow 1: 0$; gradient) gave in order of elution the ethylene glycol dispiroketal $12(42.0 \mathrm{mg}$, $79 \%$ ) and the title compound $\mathbf{1 1}(21.0 \mathrm{mg}, 62 \%)$. 11: $v_{\text {max }}$ (film)/ $\mathrm{cm}^{-1} 3407,2956,1739,1642,1441,1223,1124,1059,966,935$, $741 ; \delta_{\mathrm{H}}\left(200 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 5.85(1 \mathrm{H}$, ddd, $J 17.2,10.4,5.9$, $4-\mathrm{H}), 5.34(1 \mathrm{H}, \mathrm{dt}, J 17.2,1.5,5-\mathrm{H}), 5.25(1 \mathrm{H}, \mathrm{dt}, J 10.4,1.4$, $5-\mathrm{H}), 4.47-4.30(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}, 3-\mathrm{H}), 3.79(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 3.19(1 \mathrm{H}$, $\mathrm{br}, \mathrm{OH}), 2.70(1 \mathrm{H}, \mathrm{br}, \mathrm{OH}) ; \delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 172.5(\mathrm{C}=\mathrm{O})$, 134.7 (4-C), 117.8 (5-C), 74.0, 73.8 (2-C, 3-C), 52.5 (Me); m/z (CI) $164\left(\mathrm{MNH}_{4}{ }^{+}, 100 \%\right), 132$ (18), 118 (31) [Found $\left(\mathrm{MNH}_{4}{ }^{+}\right)$ 164.0923. $\mathrm{C}_{6} \mathrm{H}_{14} \mathrm{O}_{4} \mathrm{~N}$ requires $\left.M \mathrm{NH}_{4}, 164.0923\right]$.

## (2R,6S,7R,9R)-2,9-Bis(phenylthiomethyl)-1,8,13,16-tetraoxa-dispiro[5.0.5.4]hexadecan-14-one 4

To a solution of bi(dihydropyran) 3 ( $213 \mathrm{mg}, 0.53 \mathrm{mmol}$ ) and glycolic acid ( $76 \mathrm{mg}, 1.0 \mathrm{mmol}, 1.9$ eq.) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(15 \mathrm{ml})$ was added $\mathrm{Ph}_{3} \mathrm{P} \cdot \mathrm{HBr}(20 \mathrm{mg}, 0.06 \mathrm{mmol}, 0.1 \mathrm{eq}$.). The mixture was stirred at room temperature for 6 h during which time the mixture became yellow-orange, then the solvent was evaporated in vacuo. Purification of the residue by flash chromatography (eluent: ether-petrol 1:3) gave the title compound $\mathbf{4}(0.238 \mathrm{mg}, 0.49$ $\mathrm{mmol}, 94 \%$ ) as a colourless oil; $[a]_{\mathrm{D}}^{23}-50.3$ (c $0.35, \mathrm{CHCl}_{3}$ ); $v_{\max }($ film $) / \mathrm{cm}^{-1} 2931,1745,1585,1480,1379,1301,1209,1097$, $1050 ; \delta_{\mathrm{H}}\left(600 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.33-7.31$ ( $\left.4 \mathrm{H}, \mathrm{m}, \mathrm{Ar}\right), 7.28-7.24$ ( $4 \mathrm{H}, \mathrm{m}, \mathrm{Ar}), 7.18-7.15(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar}), 4.27(1 \mathrm{H}, \mathrm{d}, J 17.4,15-\mathrm{H})$, $4.18(1 \mathrm{H}, \mathrm{d}, J 17.4,15-\mathrm{H}), 4.16-4.13(1 \mathrm{H}, \mathrm{m}, 9-\mathrm{H}), 3.81-3.77$ ( $1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}$ ), 3.07-2.96 ( $4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{SPh}$ ), 1.96-1.26 ( $12 \mathrm{H}, \mathrm{m}$ ); $\delta_{\mathrm{C}}\left(100 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 167.0(\mathrm{C}=\mathrm{O}), 136.5,136.5,129.4,129.1$, 129.0, 128.9, 126.1, 126.0 (Ar), 103.8, 95.8 (6-C, 7-C), 71.6, 71.5 (2-C, 9-C), 59.5 (15-C), 39.5, 39.1 ( $\left.\mathrm{CH}_{2} \mathrm{SPh}\right), 29.7,29.1,27.9$, 27.4, 18.1, 17.2 (3-C, 4-C, 5-C, 10-C, 11-C, 12-C); $m / z$ (EI) 486 $\left(\mathrm{M}^{+}, 3 \%\right), 410$ (22), 123 (76), 55 (100) [Found ( $\mathrm{M}^{+}$) 486.1535. $\mathrm{C}_{26} \mathrm{H}_{30} \mathrm{O}_{5} \mathrm{~S}_{2}$ requires $\left.M, 486.1535\right]$.

## (2R,6S,7R,9R,15S)-15-[(S)-(Hydroxy)phenylmethyl]-2,9-bis(phenylthiomethyl)-1,8,13,16-tetraoxadispiro[5.0.5.4]-hexadecan-14-one 7a

Using the general procedure with 4 ( $213 \mathrm{mg}, 0.44 \mathrm{mmol}$ ) and benzaldehyde ( $0.1 \mathrm{ml}, 1.0 \mathrm{mmol}$ ), the title compound $7 \mathbf{a}$
( $218 \mathrm{mg}, 0.368 \mathrm{mmol}, 84 \%$ ) was obtained as an oil after flash chromatography (eluent: ether-petrol $1: 3 \rightarrow 1: 1$; gradient); $[\alpha]_{D}^{23}$ -51.9 (c $\left.0.37, \mathrm{CHCl}_{3}\right) ; v_{\text {max }}($ film $) / \mathrm{cm}^{-1} 3484,2946,2360,1751$, 1583, 1480, 1438, 1208, 1049, 691; $\delta_{\mathrm{H}}\left(600 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.38$ ( $2 \mathrm{H}, \mathrm{d}, J 7.5, \mathrm{Ar}$ ), $7.31-7.26$ ( $11 \mathrm{H}, \mathrm{m}, \mathrm{Ar}$ ), 7.19 ( $2 \mathrm{H}, \mathrm{t}, J 7.5$, Ar), $5.08(1 \mathrm{H}, \mathrm{d}, J 5.2, \mathrm{PhC} H), 4.22(1 \mathrm{H}, \mathrm{d}, J 5.5,15-\mathrm{H}), 4.15$ ( 1 H , dtd, J 11.6, 5.8, 2.1, 9-H), 4.04 ( $1 \mathrm{H}, \mathrm{br}, \mathrm{OH}$ ), 3.58 ( 1 H , dtd, $J 10.1,5.8,2.0,2-\mathrm{H}), 3.03-2.92\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{SPh}\right), 1.95-1.21$ $(12 \mathrm{H}, \mathrm{m}) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 167.3(\mathrm{C}=\mathrm{O}), 138.6,136.4$, $136.2,129.3,129.1,129.0,129.0,127.9,127.8,126.9,126.2$, 126.0 (Ar), 104.0, 96.4 (6-C, 7-C), 74.1, 73.8 (15-C, PhCH), 71.5, 71.4 (2-C, $9-\mathrm{C}$ ), $39.3,38.7\left(\mathrm{CH}_{2} \mathrm{SPh}\right), 29.5,29.1,28.0$, 27.2, 18.0, 17.0 (3-C, 4-C, 5-C, 10-C, 11-C, 12-C); m/z (CI) $610\left(\mathrm{MNH}_{4}^{+}, 14 \%\right), 593\left(\mathrm{MH}^{+}, 6\right), 504$ (17), 411 (100), 303 (24) [Found ( $\mathrm{MH}^{+}$) 593.2030. $\mathrm{C}_{33} \mathrm{H}_{37} \mathrm{O}_{6} \mathrm{~S}_{2}$ requires $M \mathrm{H}$, 593.2031].

## Deprotection of 7a to give methyl (2S,3S)-2,3-dihydroxy-3phenylpropionoate 10 and ( $\mathbf{2}^{\prime} S, 2^{\prime \prime} S, 6^{\prime} R, 6^{\prime \prime} R$ )-1,2-O-[ $6^{\prime}, \mathbf{6}^{\prime \prime}$ -bis(phenylthiomethyl)-3' $\mathbf{3}^{\prime \prime}, 4^{\prime}, 4^{\prime \prime}, 5^{\prime}, 5^{\prime \prime}, 6^{\prime}, 6^{\prime \prime}$-octahydro- $\mathbf{2}^{\prime}, 2^{\prime \prime}-$ bi(2H-pyran- $\mathbf{2}^{\prime}, \mathbf{2}^{\prime \prime}$-diyl)]ethane-1,2-diol 13

To a solution of $7 \mathbf{7 a}(36.0 \mathrm{mg}, 0.061 \mathrm{mmol})$ in anhydrous methanol ( 2 ml ) was added CSA ( $6.2 \mathrm{mg}, 0.0267 \mathrm{mmol}, 0.4 \mathrm{eq}$.) and ethylene glycol ( $15 \mathrm{mg}, 0.24 \mathrm{mmol}, 4.0$ eq.). The mixture was heated under reflux for 4 h . After cooling the solvent was evaporated in vacuo. Purification of the residue by flash chromatography (eluent: ether-petrol $1: 1 \rightarrow$ neat ether gradient) gave the ethylene glycol dispiroketal 13 ( $28.2 \mathrm{mg}, 0.0596 \mathrm{mmol}, 98 \%$ ) and the diol $\mathbf{1 0}(11.9 \mathrm{mg}, 0.0606 \mathrm{mmol}, 99 \%)$. The absolute stereochemistry of $\mathbf{1 0}$ was determined as $(2 S, 3 S)$ by comparing the optical rotation with that reported in the literature \{ref. $[a]_{\mathrm{D}}^{22}$ $-41.3\left(c 0.48, \mathrm{CHCl}_{3}\right)$ for the $(2 R, 3 R)$-form $\} .^{12}$ All spectroscopic data of enantiopure $\mathbf{1 0}$ were identical to racemic $\mathbf{1 0}$ (see above).

10: $[a]_{\mathrm{D}}^{23}+36.1\left(c 0.72, \mathrm{CHCl}_{3}\right)$.
13: $[a]_{\mathrm{D}}^{24}-71.7\left(c 0.90, \mathrm{CHCl}_{3}\right) ; v_{\max }($ film $) / \mathrm{cm}^{-1} 2926,1630$, 1584, 1481, 1438, 1287, 1082, 1036, 966, 911, 736, 690; $\delta_{\mathrm{H}}(600$ $\mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) $7.37(4 \mathrm{H}, \mathrm{d}, J 7.4, \mathrm{Ar}), 7.26(4 \mathrm{H}, \mathrm{t}, J 7.7, \mathrm{Ar})$, $7.15(2 \mathrm{H}, \mathrm{t}, J 7.4, \mathrm{Ar}), 4.01\left(2 \mathrm{H}, \mathrm{d}, J 7.6, \mathrm{CH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}\right), 3.89-3.84$ $\left(2 \mathrm{H}, \mathrm{m}, 6^{\prime}-\mathrm{H}, 6^{\prime \prime}-\mathrm{H}\right), 3.38\left(2 \mathrm{H}, \mathrm{d}, J 7.7, \mathrm{CH}_{\mathrm{A}} H_{\mathrm{B}}\right), 3.13(2 \mathrm{H}, \mathrm{dd}$, $\left.J 13.3,6.2, \mathrm{C}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}} \mathrm{SPh}\right), 2.95\left(2 \mathrm{H}, \mathrm{dd}, J 13.3,6.0, \mathrm{CH}_{\mathrm{A}} H_{\mathrm{B}} \mathrm{SPh}\right)$, $1.82-1.69(6 \mathrm{H}, \mathrm{m}), 1.62-1.57(2 \mathrm{H}, \mathrm{m}), 1.43(2 \mathrm{H}, \mathrm{td}, J 13.4,4.8)$, $1.26-1.20(2 \mathrm{H}, \mathrm{m}) ; \delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 137.2,128.9,128.7$, 125.7 (Ar), 96.6 ( $\left.2^{\prime}-\mathrm{C}, 2^{\prime \prime}-\mathrm{C}\right), 69.4$ ( $6^{\prime}-\mathrm{C}, 6^{\prime \prime}-\mathrm{C}$ ), 58.3 ( $1-\mathrm{C}, 2-\mathrm{C}$ ), $39.4\left(\mathrm{CH}_{2} \mathrm{SPh}\right)$, 29.9, 28.0, 18.1 ( $3^{\prime}-\mathrm{C}, 4^{\prime}-\mathrm{C}, 5^{\prime}-\mathrm{C}, 3^{\prime \prime}-\mathrm{C}, 4^{\prime \prime}-\mathrm{C}$, $\left.5^{\prime \prime-} \mathrm{C}\right) ; m / z(\mathrm{CI}) 490\left(\mathrm{MNH}_{4}{ }^{+}, 46 \%\right), 473\left(\mathrm{MH}^{+}, 12\right), 411$ (100), 303 (53) [Found ( $\mathrm{MH}^{+}$) 473.1820. $\mathrm{C}_{26} \mathrm{H}_{33} \mathrm{O}_{4} \mathrm{~S}_{2}$ requires $M \mathrm{H}$, 473.1820].

## Mosher esterification of methyl (2S,3S)-2,3-dihydroxy-3-phenylpropionoate 10 prepared above

To a stirred solution of methyl ( $2 S, 3 S$ )-2,3-dihydroxy-3-phenylpropionoate $10(14.4 \mathrm{mg}, 0.073 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1 \mathrm{ml})$, was added $\mathrm{Et}_{3} \mathrm{~N}(0.1 \mathrm{ml})$, DMAP ( 5 mg ) and $(R)-(-)$-MTPACl ( $\alpha$-methoxy- $\alpha$-trifluoromethylphenylacetyl chloride, 100 ml ) at $0{ }^{\circ} \mathrm{C}$, and the resulting mixture was allowed to warm up to room temperature over 1 h . The mixture was quenched by water and extracted with ether $(\times 3)$. The combined organic phases were washed with brine, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporated in vacuo under reduced pressure. The resulting residue was purified by flash chromatography (eluent: ether-petrol 1:1) to give the bis(MTPA) ester ( $62.6 \mathrm{mg}, 85 \%$ ) as an oil; $\delta_{\mathrm{H}}(200 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 7.50-7.00(15 \mathrm{H}, \mathrm{m}, \mathrm{Ar}), 6.39(1 \mathrm{H}, \mathrm{d}, J 4.5,2-\mathrm{H}$ or $3-\mathrm{H})$, $5.70(1 \mathrm{H}, \mathrm{d}, J 4.5,2-\mathrm{H}$ or $3-\mathrm{H}), 3.73(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 3.46(3 \mathrm{H}, \mathrm{d}$, $J 1.2, \mathrm{Me}), 3.43(3 \mathrm{H}, \mathrm{d}, J 1.2, \mathrm{Me}) ; \delta_{\mathrm{F}}\left(235 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $-72.13\left(3 \mathrm{~F}, \mathrm{~s}, \mathrm{CF}_{3}\right),-72.36\left(3 \mathrm{~F}, \mathrm{~s}, \mathrm{CF}_{3}\right)$.
The other isomer was not detected with ${ }^{1} \mathrm{H}$ and ${ }^{19} \mathrm{~F}$ NMR, which implies the optical purity of the initial diol is $>95 \%$ ee.
Data for the other isomer obtained from the racemic sample;
$\delta_{\mathrm{H}}\left(200 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.50-7.00(15 \mathrm{H}, \mathrm{m}, \mathrm{Ar}), 6.45(1 \mathrm{H}, \mathrm{d}$, $J 5.4,2-\mathrm{H}$ or $3-\mathrm{H}), 5.62(1 \mathrm{H}, \mathrm{d}, J 5.4,2-\mathrm{H}$ or $3-\mathrm{H}), 3.62(3 \mathrm{H}, \mathrm{s}$, $\mathrm{Me}), 3.41(3 \mathrm{H}, \mathrm{d}, J 1.2, \mathrm{Me}), 3.24(3 \mathrm{H}, \mathrm{d}, J 1.2, \mathrm{Me}) ; \delta_{\mathrm{F}}(235$ $\left.\mathrm{MHz} ; \mathrm{CDCl}_{3}\right)-71.41\left(3 \mathrm{~F}, \mathrm{~s}, \mathrm{CF}_{3}\right),-72.41\left(3 \mathrm{~F}, \mathrm{~s}, \mathrm{CF}_{3}\right)$.
( $1^{\prime} S, 2 R, 6 S, 7 R, 9 R, 15 S$ )-15-(1-Hydroxyprop-2-enyl)-2,9-bis(phenylthiomethyl)-1,8,13,16-tetraoxadispiro[5.0.5.4]hexa-decan-14-one 8 a and ( $1^{\prime} R, 2 R, 6 S, 7 R, 9 R, 15 R$ )-15-(1-hydroxy-prop-2-enyl)-2,9-bis(phenylthiomethyl)-1,8,13,16-tetraoxa-dispiro[5.0.5.4]hexadecan-14-one 8b

The general procedure using $4(76.9 \mathrm{mg}, 0.158 \mathrm{mmol})$ and acrolein ( $25 \mathrm{ml}, 0.374 \mathrm{mmol}, 2.0$ eq.) gave in order of elution $\mathbf{8 b}$ $(2.4 \mathrm{mg}, 0.004 \mathrm{mmol}, 3 \%)$ and $\mathbf{8 a}(54.6 \mathrm{mg}, 0.10 \mathrm{mmol}, 64 \%)$ following flash chromatography (eluent: ether-petrol $1: 3 \rightarrow 1: 1$; gradient).

8a: $[a]_{\mathrm{D}}^{21}-79.1\left(c 0.69, \mathrm{CHCl}_{3}\right) ; v_{\max }(\mathrm{film}) / \mathrm{cm}^{-1} 3495,2952$, $1748,1584,1481,1439,1208,1041 ; \delta_{\mathrm{H}}\left(600 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.36$ (2H, d, J7.8, Ar), 7.31 (2H, d, J8.1, Ar), 7.28-7.24 (4H, m, Ar), 7.18-7.15 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{Ar}$ ), $6.00\left(1 \mathrm{H}\right.$, ddd, $\left.J 17.2,10.5,6.2,2^{\prime}-\mathrm{H}\right)$, $5.38\left(1 \mathrm{H}, \mathrm{dt}, J 17.2,1.2,3^{\prime}-\mathrm{H}\right), 5.25\left(1 \mathrm{H}, \mathrm{dt}, J 10.5,1.2,3^{\prime}-\mathrm{H}\right)$, 4.51-4.47 ( $\left.1 \mathrm{H}, \mathrm{m}, 1^{\prime}-\mathrm{H}\right), 4.16(1 \mathrm{H}, \mathrm{dtd}, J 11.7,5.7,2.4,9-\mathrm{H})$, $4.14(1 \mathrm{H}, \mathrm{d}, J 3.8,15-\mathrm{H}), 3.76$ ( 1 H , dtd, $J 11.7,5.9,1.9,2-\mathrm{H}$ ), $3.39(1 \mathrm{H}, \mathrm{d}, J 2.3, \mathrm{OH}), 3.05-2.96\left(4 \mathrm{H}, \mathrm{m}, \mathrm{C} H_{2} \mathrm{SPh}\right), 1.97-1.26$ $(12 \mathrm{H}, \mathrm{m}) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 166.4(\mathrm{C}=\mathrm{O}), 136.3,136.3$, 134.9, 129.4, 129.1, 129.0, 128.9, 126.2, 126.1 ( $\mathrm{Ar}, 2^{\prime}$-C), 117.4 ( $3^{\prime}-\mathrm{C}$ ), 103.8, 96.4 (6-C, 7-C), 73.9, 73.6 (15-C, 1'-C), 71.5, 71.4 (2-C, 9-C), 39.4, $38.9\left(\mathrm{CH}_{2} \mathrm{SPh}\right), 29.5,29.1,28.2,27.1,18.1$, 17.0 (3-C, $4-\mathrm{C}, 5-\mathrm{C}, 10-\mathrm{C}, 11-\mathrm{C}, 12-\mathrm{C}) ; ~ m / z(\mathrm{CI}) 560\left(\mathrm{MNH}_{4}{ }^{+}\right.$, $44 \%), 543\left(\mathrm{MH}^{+}, 17\right), 504$ (24), 411 (100) [Found ( $\mathrm{MH}^{+}$) 543.1875. $\mathrm{C}_{29} \mathrm{H}_{35} \mathrm{O}_{6} \mathrm{~S}_{2}$ requires $M \mathrm{H}, 543.1875$ ].

8b: $[a]_{\mathrm{D}}^{26}-55.5\left(c 0.51, \mathrm{CHCl}_{3}\right) ; v_{\max }($ film $) / \mathrm{cm}^{-1} 3488,2920$, 1731, 1654, 1584, 1481, 1439, 1206, 1041, 736; $\delta_{\mathrm{H}}(600 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 7.40-7.14(10 \mathrm{H}, \mathrm{m}, \mathrm{Ar}), 5.94(1 \mathrm{H}, \mathrm{ddd}, J 17.3,10.6,5.6$, $\left.2^{\prime}-\mathrm{H}\right), 5.24\left(1 \mathrm{H}, \mathrm{d}, J 17.3,3^{\prime}-\mathrm{H}\right), 5.20\left(1 \mathrm{H}, \mathrm{d}, J 10.6,3^{\prime}-\mathrm{H}\right)$, 4.30-3.95 ( $\left.4 \mathrm{H}, \mathrm{m}, 1^{\prime}-\mathrm{H}, 2-\mathrm{H}, 9-\mathrm{H}, 15-\mathrm{H}\right), 3.81(1 \mathrm{H}, \mathrm{d}, J 3.0$, $\mathrm{OH}), 3.04-2.97\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{SPh}\right), 1.95-1.25(12 \mathrm{H}, \mathrm{m})$; $\delta_{\mathrm{C}}(50$ $\mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) $171.0(\mathrm{C}=\mathrm{O}), 136.4,136.2,135.7,129.3,128.9$, 126.1, 125.9 (Ar, 2'-C), 117.3 (3'-C), 104.4, 96.7 (6-C, 7-C), 72.3, 72.3, 72.2, 70.6 ( $1^{\prime}-\mathrm{C}, 2-\mathrm{C}, ~ 9-\mathrm{C}, 15-\mathrm{C}$ ), $39.3,38.9$ $\left(\mathrm{CH}_{2} \mathrm{SPh}\right), 29.6,29.3,28.3,27.7,18.0,17.5$ (3-C, 4-C, 5-C, 10-C, $11-\mathrm{C}, 12-\mathrm{C}) ; m / z$ (CI) $560\left(\mathrm{MNH}_{4}^{+}, 3 \%\right), 543\left(\mathrm{MH}^{+}, 2\right), 411$ (9), 303 (11), 78 (100) [Found $\left(\mathrm{MH}^{+}\right) 543.1875 . \mathrm{C}_{29} \mathrm{H}_{35} \mathrm{O}_{6} \mathrm{~S}_{2}$ requires $M \mathrm{H}, 543.1875]$.

## Deprotection of 8a to give methyl (2S,3S)-2,3-dihydroxypent-4enoate 11

To a solution of $8 \mathbf{~ ( ~} 54.3 \mathrm{mg}, 0.10 \mathrm{mmol}$ ) in anhydrous methanol ( 2.5 ml ) was added CSA ( $8 \mathrm{mg}, 0.03 \mathrm{mmol}, 0.3$ eq.) and ethylene glycol ( $15 \mathrm{mg}, 0.24 \mathrm{mmol}, 2.4$ eq.). The mixture was heated under reflux for 4 h . After cooling the solvent was evaporated in vacuo. Purification of the residue by flash chromatography (eluent: ether-petrol $1: 1 \rightarrow$ neat ether; gradient) gave the ethylene glycol dispiroketal $13(45.3 \mathrm{mg}, 0.096 \mathrm{mmol}, 96 \%)$ and $\mathbf{1 1}(11.9 \mathrm{mg}, 0.081 \mathrm{mmol}, 81 \%)$. 11: $[a]_{\mathrm{D}}^{23}+2.7$ (c 1.19 , $\mathrm{CDCl}_{3}$.

## Mosher esterification of methyl (2S,3S)-2,3-dihydroxypent-4enoate 11 prepared in above

To a stirred solution of methyl ( $2 S, 3 S$ )-2,3-dihydroxypent-4enoate $11(11.9 \mathrm{mg}, 0.081 \mathrm{mmol}), \mathrm{Et}_{3} \mathrm{~N}(0.1 \mathrm{ml})$ and DMAP ( 5 $\mathrm{mg})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1.0 \mathrm{ml})$ was added $(R)-(-)-\mathrm{MTPACl}(100 \mathrm{ml})$ at $0^{\circ} \mathrm{C}$, and the mixture was allowed to warm up to room temperature over 1 h . The mixture was quenched by water and extracted with ether $(\times 3)$. The combined organic extracts were washed with brine, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporated in vacuo. The resulting residue was purified by flash chromatography (eluent: ether-petrol $1: 1$ ) to give the bis(MTPA)ester $(22.7 \mathrm{mg}$, $0.0392 \mathrm{mmol}, 48 \%)$ as an oil; $\delta_{\mathrm{H}}\left(600 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.60(2 \mathrm{H}, \mathrm{d}$, $J 7.8, \mathrm{Ar}), 7.44(2 \mathrm{H}, \mathrm{d}, J 7.7, \mathrm{Ar}), 7.41-7.32(6 \mathrm{H}, \mathrm{m}, \mathrm{Ar}), 5.90$
(1H, dd, $J 7.5,2.8,3-\mathrm{H}), 5.73$ (1H, ddd, $J 17.2,10.5,7.5,4-\mathrm{H})$, $5.60(1 \mathrm{H}, \mathrm{d}, J 3.0,2-\mathrm{H}), 5.35(1 \mathrm{H}, \mathrm{d}, J 19.4,5-\mathrm{H}), 3.44(1 \mathrm{H}, \mathrm{d}$, $J 10.7,5-\mathrm{H}), 3.78(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 3.54(3 \mathrm{H}, \mathrm{d}, J 1.2, \mathrm{Me}), 3.44(3 \mathrm{H}$, d, $J 1.2, \mathrm{Me}) ; \delta_{\mathrm{F}}\left(235 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)-72.19\left(3 \mathrm{~F}, \mathrm{~s}, \mathrm{CF}_{3}\right),-72.39$ (3F, s, $\mathrm{CF}_{3}$ ).

The other isomer was not detected by ${ }^{1} \mathrm{H}$ or ${ }^{19} \mathrm{~F}$ NMR spectroscopy, which implies that the optical purity of the initial diol is $>95 \%$ ee.

Data for the other isomer obtained from the racemic sample; $\delta_{\mathrm{H}}\left(200 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.62-7.28(10 \mathrm{H}, \mathrm{m}, \mathrm{Ar}), 6.03-5.40(5 \mathrm{H}$, m), $3.71(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 3.48(3 \mathrm{H}, \mathrm{d}, J 1.2, \mathrm{Me}), 3.43(3 \mathrm{H}, \mathrm{d}, J 1.2$, $\mathrm{Me}) ; \delta_{\mathrm{F}}\left(235 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)-72.15\left(3 \mathrm{~F}, \mathrm{~s}, \mathrm{CF}_{3}\right),-72.38(3 \mathrm{~F}, \mathrm{~s}$, $\mathrm{CF}_{3}$ ).

## ( $\left.1^{\prime} S, 2 R, 2^{\prime} R, 6 S, 7 R, 9 R, 15 S\right)-15-\left[\left(2^{\prime}, 3^{\prime}-O\right.\right.$-Isopropylidene-1' $\mathbf{2}^{\prime}$, $3^{\prime}$-trihydroxy)propyl]-2,9-bis(phenylthiomethyl)-1,8,13,16-tetra-oxadispiro[5.0.5.4]hexadecan-14-one 15

Using the general procedure with $\mathbf{4}(86.2 \mathrm{mg}, 0.177 \mathrm{mmol})$ and $\mathbf{1 4}(46 \mathrm{mg}, 0.346 \mathrm{mmol}, 2.0 \mathrm{eq}$.$) , a mixture of 15$ and $\mathbf{1 6}$ ( 15.6 $\mathrm{mg}, 0.0252 \mathrm{mmol}, 14 \%$ ) was obtained after flash chromatography (eluent: ether-petrol $1: 3 \rightarrow 1: 1$; gradient). The ratio of 15 and 16 was determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy as $6: 4$.

15: $\delta_{\mathrm{H}}\left(600 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.35-7.12(10 \mathrm{H}, \mathrm{m}, \mathrm{Ar}), 4.68(1 \mathrm{H}$, d, $J 4.8,15-\mathrm{H}), 4.68\left(1 \mathrm{H}, \mathrm{q}, J 6.6,2^{\prime}-\mathrm{H}\right), 4.29-4.25(1 \mathrm{H}, \mathrm{m}, 9-\mathrm{H}$ or $2-\mathrm{H}), 4.10-3.95\left(3 \mathrm{H}, \mathrm{m}, 3^{\prime}-\mathrm{H} \times 2,2-\mathrm{H}\right.$ or $\left.9-\mathrm{H}\right), 3.84-3.80$ $\left(1 \mathrm{H}, \mathrm{m}, 1^{\prime}-\mathrm{H}\right), 3.19(1 \mathrm{H}, \mathrm{br}, \mathrm{OH}), 3.03-2.95\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{SPh}\right)$, $1.95-1.20(12 \mathrm{H}, \mathrm{m}), 1.40(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 1.35(3 \mathrm{H}, \mathrm{s}, \mathrm{Me})$.

16: $\delta_{\mathrm{H}}\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.35-7.12(10 \mathrm{H}, \mathrm{m}, \mathrm{Ar}), 4.32(1 \mathrm{H}$, q, $\left.J 6.9,2^{\prime}-\mathrm{H}\right), 4.19-4.15(1 \mathrm{H}, \mathrm{m}, 9-\mathrm{H}$ or $2-\mathrm{H}), 4.10-3.95(3 \mathrm{H}$, $\left.\mathrm{m}, 3^{\prime}-\mathrm{H} \times 2,15-\mathrm{H}\right), 3.80-3.78(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}$ or $9-\mathrm{H}), 3.67(1 \mathrm{H}, \mathrm{t}$, $\left.J 7.8,1^{\prime}-\mathrm{H}\right), 3.49(1 \mathrm{H}, \mathrm{br}, \mathrm{OH}), 3.03-2.95\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{SPh}\right)$, $1.95-1.20(12 \mathrm{H}, \mathrm{m}), 1.43$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{Me}$ ), 1.39 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{Me}$ ).

## (2S,6R,7S,9S)-2,9-Bis(phenylthiomethyl)-1,8,13,16-tetraoxa-dispiro[5.0.5.4]hexadecan-14-one 18

To a solution of bi(dihydropyran) $\mathbf{1 7}(0.84 \mathrm{~g}, 2.05 \mathrm{mmol})$ and glycolic acid ( $312 \mathrm{mg}, 4.1 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(60 \mathrm{ml})$ was added $\mathrm{Ph}_{3} \mathrm{P} \cdot \mathrm{HBr}(68.6 \mathrm{mg}, 0.20 \mathrm{mmol})$. The mixture was stirred at room temperature for 8 h during which time the mixture became yellow-orange, then the solvent was evaporated in vacuo. Purification of the residue by flash chromatography (eluent: ether-petrol $1: 3$ ) gave the title compound $\mathbf{1 8}(0.86 \mathrm{mg}$, $1.77 \mathrm{mmol}, 86 \%)$.
( $\left.1^{\prime} R, 2 S, 2^{\prime} R, 6 R, 7 S, 9 S, 15 R\right)-15-\left[\left(2^{\prime}, 3^{\prime}-O\right.\right.$-Isopropylidene- $1^{\prime}, 2^{\prime}$, $3^{\prime}$-trihydroxy)propyl]-2,9-bis(phenylthiomethyl)-1,8,13,16-tetraoxadispiro[5.0.5.4] hexadecan-14-one 19 and $\left(1^{\prime} S, 2 S, 2^{\prime} R, 6 R\right.$, $7 S, 9 S, 15 R)$-15-[( $\mathbf{2}^{\prime}, \mathbf{3}^{\prime}$ - O-isopropylidene-1 $\mathbf{1}^{\prime}, \mathbf{2}^{\prime}, \mathbf{3}^{\prime}$-trihydroxy)-propyl]-2,9-bis(phenylthiomethyl)-1,8,13,16-tetraoxadispiro-[5.0.5.4]hexadecan-14-one 20
The general procedure using $\mathbf{1 8}(103 \mathrm{mg}, 0.212 \mathrm{mmol})$ and $\mathbf{1 4}$ $(55 \mathrm{mg}, 0.424 \mathrm{mmol}, 2.0$ eq.) gave in order of elution the reactant $18(14.1 \mathrm{mg}, 0.0289 \mathrm{mmol}, 14 \%)$, the mixture of the two most minor diastereomers ( $4.7 \mathrm{mg}, 0.0076 \mathrm{mmol}, 4 \%$ ), $\mathbf{2 0}$ ( 11.5 $\mathrm{mg}, 0.086 \mathrm{mmol}, 19 \%$ ), and 19 ( $70.6 \mathrm{mg}, 0.1144 \mathrm{mmol}, 54 \%$ ) after chromatography (eluent: ether-petrol $1: 3 \rightarrow 1: 1$; gradient).

19: $[a]_{\mathrm{D}}^{23}+56.0\left(c 1.41, \mathrm{CDCl}_{3}\right) ; v_{\max }($ film $) / \mathrm{cm}^{-1} 3481,2935$, 2937, 1748, 1584, 1481, 1439, 1372, 1302, 1254, 1209, 1161, 1106, 1043, 983, 956, 913, 844, 731, 691; $\delta_{\mathrm{H}}\left(600 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right.$ ) 7.36-7.14 ( $10 \mathrm{H}, \mathrm{m}, \mathrm{Ar}$ ), $4.46(1 \mathrm{H}, \mathrm{d}, J 1.6,15-\mathrm{H}), 4.38(1 \mathrm{H}, \mathrm{q}$, $\left.J 6.5,2^{\prime}-\mathrm{H}\right), 4.18-4.09\left(2 \mathrm{H}, \mathrm{m}, 3^{\prime}-\mathrm{H}, 9-\mathrm{H}\right), 4.04-4.00(1 \mathrm{H}, \mathrm{m}$, $\left.3^{\prime}-\mathrm{H}\right), 3.95\left(1 \mathrm{H}, \mathrm{d}, J 8.0,1^{\prime}-\mathrm{H}\right), 3.82-3.79(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}), 3.27$ $(1 \mathrm{H}, \mathrm{d}, J 2.2, \mathrm{OH}), 3.05-2.98\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{SPh}\right), 1.96-1.65(6 \mathrm{H}$, $\mathrm{m}), 1.53-1.24(6 \mathrm{H}, \mathrm{m}), 1.40(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 1.38(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}) ; \delta_{\mathrm{C}}(50$ $\left.\mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 165.9(\mathrm{C}=\mathrm{O}), 136.4,136.1,129.4,128.9,128.9$, 128.8, 126.2, 125.8 (Ar), 109.2, 103.4, 96.2 (6-C, 7-C, $\mathrm{Me}_{2} \mathrm{C}$ ), 74.3, 74.0, 72.1, 71.4, 71.4 (1'-C, 2-C, $\left.2^{\prime}-\mathrm{C}, 9-\mathrm{C}, 15-\mathrm{C}\right), 66.8$ (3'-C), $39.6\left(\mathrm{CH}_{2} \mathrm{SPh}\right), 38.8\left(\mathrm{CH}_{2} \mathrm{SPh}\right), 29.3,29.2,28.1,27.2$, 18.0, 16.9 (3-C, 4-C, 5-C, 10-C, 11-C, 12-C), 26.8 (Me), 25.4
(Me); $m / z$ (CI) $634\left(\mathrm{MNH}_{4}^{+}, 14 \%\right), 617\left(\mathrm{MH}^{+}, 16\right), 411(42)$, 303 (100), 240 (39) [Found ( $\mathrm{MH}^{+}$) 617.2240. $\mathrm{C}_{32} \mathrm{H}_{41} \mathrm{O}_{8} \mathrm{~S}_{2}$ requires $M \mathrm{H}, 617.2243]$.
20: $[a]_{\mathrm{D}}^{23}+49.9\left(c 1.15, \mathrm{CDCl}_{3}\right) ; v_{\max }(\mathrm{film}) / \mathrm{cm}^{-1} 3493,2984$, 2938, 1749, 1584, 1481, 1439, 1372, 1302, 1256, 1209, 1160, $1144,1044,982,912,852,735,691 ; \delta_{\mathrm{H}}\left(600 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.36-$ $7.13(10 \mathrm{H}, \mathrm{m}, \mathrm{Ar}), 4.23-4.19\left(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}, 2^{\prime}-\mathrm{H}\right), 4.11(1 \mathrm{H}, \mathrm{d}$, $J 1.6,15-\mathrm{H}), 4.03-3.99\left(1 \mathrm{H}, \mathrm{m}, 1^{\prime}-\mathrm{H}\right), 3.80\left(1 \mathrm{H}, \mathrm{t}, J 7.7,3^{\prime}-\mathrm{H}\right)$, 3.69-3.64 ( $1 \mathrm{H}, \mathrm{m}, 9-\mathrm{H}$ ), $3.59\left(1 \mathrm{H}, \mathrm{t}, J 7.7,3^{\prime}-\mathrm{H}\right), 3.15(1 \mathrm{H}, \mathrm{d}$, $J 10.4, \mathrm{OH}), 3.03\left(2 \mathrm{H}, \mathrm{d}, J 5.6, \mathrm{CH}_{2} \mathrm{SPh}\right), 3.00(2 \mathrm{H}, \mathrm{d}, J 6.0$, $\left.\mathrm{CH}_{2} \mathrm{SPh}\right), 2.00-1.67(8 \mathrm{H}, \mathrm{m}), 1.60-1.20(4 \mathrm{H}, \mathrm{m}), 1.40(3 \mathrm{H}, \mathrm{s}$, Me ), $1.36(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}) ; \delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 167.6(\mathrm{C}=\mathrm{O}), 136.4$, 136.1, 129.2, 129.0, 128.9, 128.8, 126.1, 126.0 (Ar), 109.7, 103.4, 96.2 (6-C, 7-C, $\mathrm{Me}_{2} \mathrm{C}$ ), 76.4, 72.8, 71.6, 71.3, 70.7 (1'-C, 2-C, $\left.2^{\prime}-\mathrm{C}, 9-\mathrm{C}, 15-\mathrm{C}\right), 65.7\left(3^{\prime}-\mathrm{C}\right), 39.1\left(\mathrm{CH}_{2} \mathrm{SPh}\right), 38.6\left(\mathrm{CH}_{2} \mathrm{SPh}\right)$, 29.5, 28.7, 27.7, 27.1, 18.1, 16.8 (3-C, 4-C, 5-C, 10-C, 11-C, $12-\mathrm{C}), 26.4(\mathrm{Me}), 25.6(\mathrm{Me}) ; \mathrm{m} / \mathrm{z}(\mathrm{CI}) 634\left(\mathrm{MNH}_{4}{ }^{+}, 34 \%\right)$, $617\left(\mathrm{MH}^{+}, 21\right), 411$ (96), 303 (100), 240 (37) [Found ( $\mathrm{MH}^{+}$) 617.2240. $\mathrm{C}_{32} \mathrm{H}_{41} \mathrm{O}_{8} \mathrm{~S}_{2}$ requires $\mathrm{MH}, 617.2243$ ].

## ( $\left.6 R^{*}, 7 R^{*}, 15 S^{*}\right)-15-\left[\left(S^{*}\right)\right.$-(tert-Butyldimethylsilyloxy)phenyl-methyl]-1,8,13,16-tetraoxadispiro[5.0.5.4]hexadecan-14-one 21

To a solution of 5 a ( $163 \mathrm{mg}, 0.47 \mathrm{mmol}$ ) in DMF ( 2.5 ml ), imidazole ( $106 \mathrm{mg}, 1.56 \mathrm{mmol}, 3.3$ eq.) and tert-butyldimethylsilyl chloride ( $122 \mathrm{mg}, 0.81 \mathrm{mmol}, 1.7 \mathrm{eq}$.) were added and stirred at room temperature for 1 day. Due to incomplete reaction, extra imidazole ( $122 \mathrm{mg}, 1.79 \mathrm{mmol}, 3.8 \mathrm{eq}$.) and tert-butyldimethylsilyl chloride ( $128 \mathrm{mg}, 0.85 \mathrm{mmol}, 1.8$ eq.) were added to the mixture and stirred for another 2 days. The reaction was then quenched with $\mathrm{H}_{2} \mathrm{O}$ and the mixture was extracted with ether ( $\times 3$ ). The combined organic extracts were dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated in vacuo. Flash chromatography (eluent: ether-petrol 1:7) gave the title compound 21 as a colourless oil ( $214 \mathrm{mg}, 0.46 \mathrm{mmol}, 98 \%$ ); $v_{\max }$ (film) $/ \mathrm{cm}^{-1} 2951,2856$, 1742, 1471, 1356, 1253, 1102, 1074, 1002, 952, 836, 778; $\delta_{\mathrm{H}}(600$ $\left.\mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.38$ ( $2 \mathrm{H}, \mathrm{d}, J 7.2, \mathrm{Ar}$ ), 7.24 ( $2 \mathrm{H}, \mathrm{t}, J 7.3, \mathrm{Ar}$ ), $7.20(1 \mathrm{H}, \mathrm{t}, J 7.2, \mathrm{Ar}), 5.26(1 \mathrm{H}, \mathrm{d}, J 3.5, \mathrm{PhCH}), 4.44(1 \mathrm{H}, \mathrm{d}$, $J 3.6,15-\mathrm{H}), 3.75-3.71(1 \mathrm{H}, \mathrm{m}), 3.64-3.59(1 \mathrm{H}, \mathrm{m}), 3.28-3.22$ $(1 \mathrm{H}, \mathrm{m}), 3.13-3.09(1 \mathrm{H}, \mathrm{m}), 1.86-1.36(12 \mathrm{H}, \mathrm{m}), 0.91(9 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{Bu}^{t}\right), 0.10(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}),-0.08(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}) ; \delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ 167.4 (C=O), 139.9, 127.9, 127.2, 127.1 (Ar), 103.2, 95.2 (6-C, 7-C), 75.1, 74.5 ( $15-\mathrm{C}, \mathrm{PhCH}$ ), 62.2, 61.9 (2-C, 9-C), 28.3, 27.8, 24.7, 24.1, 18.1, 17.1 (3-C, 4-C, 5-C, 10-C, 11-C, 12-C), 25.7 $\left(\mathrm{Bu}^{t}\right), 18.1\left(\mathrm{Bu}^{t}\right),-4.9(\mathrm{Me}),-5.1(\mathrm{Me}) ; m / z(\mathrm{CI}) 463\left(\mathrm{MH}^{+}\right.$, $3 \%$ ), 167 (100) [Found $\left(\mathrm{MH}^{+}\right) 463.2516 . \mathrm{C}_{25} \mathrm{H}_{39} \mathrm{O}_{6} \mathrm{Si}$ requires $M \mathrm{H}, 463.2516]$.
( $6 R^{*}, 7 R^{*}, 14 S^{*}$ )-14-[( $\left.S^{*}\right)$-(tert-Butyldimethylsilyloxy)phenyl-methyl]-15-methylene-1,8,13,16-tetraoxadispiro[5.0.5.4]hexadecane 23
To a solution of $\mathbf{2 1}$ ( $213.5 \mathrm{mg}, 0.46 \mathrm{mmol}$ ) in THF ( 5 ml ) and pyridine ( 1 ml ) was added a 0.5 M solution of the Tebbe reagent in toluene ( $1.4 \mathrm{ml}, 0.69 \mathrm{mmol}, 1.5 \mathrm{eq}$.) at $-78^{\circ} \mathrm{C}$. After stirring for 10 min , the mixture was warmed up to room temperature and stirred for 1 h . Then, the mixture was quenched with a saturated $\mathrm{Na}_{2} \mathrm{CO}_{3}$ solution ( 1 ml ) and ether at $0^{\circ} \mathrm{C}$. The solid formed was filtered over Celite and washed with ether. The organic filtrate was dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated in vauco. The resulting residue was purified by flash chromatography (eluent: ether-petrol 1:7) to give the title compound 23 (204 $\mathrm{mg}, 0.443 \mathrm{mmol}, 96 \%$ ); mp 109-110 ${ }^{\circ} \mathrm{C}$ (ether-petrol) (Found: $\mathrm{C}, 67.94 ; \mathrm{H}, 8.74 . \mathrm{C}_{26} \mathrm{H}_{40} \mathrm{O}_{5} \mathrm{Si}$ requires $\mathrm{C}, 67.79 ; \mathrm{H}, 8.75 \%$ ); $v_{\max }($ film $) / \mathrm{cm}^{-1} 2951,2856,1742,1471,1356,1253,1102,1074$, 1002, 952, 836, 778; $\delta_{\mathrm{H}}\left(600 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.42(2 \mathrm{H}, \mathrm{d}, J 7.3$, Ar), $7.28(2 \mathrm{H}, \mathrm{t}, J 7.3, \mathrm{Ar}), 7.22(1 \mathrm{H}, \mathrm{t}, J 7.3, \mathrm{Ar}), 4.88(1 \mathrm{H}, \mathrm{d}$, $J 8.2, \mathrm{PhCH}), 4.79\left(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2}=\right), 4.74\left(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2}=\right), 4.22(1 \mathrm{H}$, d, $J 8.2,14-\mathrm{H}), 4.00-3.93(1 \mathrm{H}, \mathrm{m}), 3.72-3.68(1 \mathrm{H}, \mathrm{m}), 3.44-3.40$ $(1 \mathrm{H}, \mathrm{m}), 2.93(1 \mathrm{H}, \mathrm{td}, J 10.8,1.9), 1.95-1.87(1 \mathrm{H}, \mathrm{m}), 1.86-1.81$ $(1 \mathrm{H}, \mathrm{m}), 1.64-1.57(3 \mathrm{H}, \mathrm{m}), 1.55-1.50(2 \mathrm{H}, \mathrm{m}), 1.46(1 \mathrm{H}, \mathrm{td}$,
$J 13.5,4.7), 1.36-1.25(2 \mathrm{H}, \mathrm{m}), 1.07-1.03(2 \mathrm{H}, \mathrm{m}), 0.87(9 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{Bu}^{t}\right), 0.09(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}),-0.37(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}) ; \delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ 153.4 (14-C), 142.4, 127.7, 127.4, 127.3 (Ar), 99.1, 96.6 (6-C, 7-C), $95.4\left(\mathrm{CH}_{2}=\right), 74.5,70.6(14-\mathrm{C}, \mathrm{PhCH}), 61.0,60.7$ (2-C, 9-C), 28.4, 28.3, 24.8, 24.6, 17.9, 17.5 (3-C, 4-C, 5-C, 10-C, $11-\mathrm{C}, 12-\mathrm{C}), 29.6\left(\mathrm{Bu}^{t}\right), 17.9\left(\mathrm{Bu}^{t}\right),-4.0(\mathrm{Me}),-5.1(\mathrm{Me}) ; m / z$ (CI) $461\left(\mathrm{MH}^{+}, 3 \%\right), 329$ (63), 167 (100) [Found $\left(\mathrm{MH}^{+}\right)$ 461.2723. $\mathrm{C}_{26} \mathrm{H}_{41} \mathrm{O}_{5} \mathrm{Si}$ requires $M \mathrm{H}, 461.2723$ ].
( $\left.6 R^{*}, 7 R^{*}, 14 R^{*}, 15 R^{*}\right)-14-\left[\left(S^{*}\right)\right.$-(tert-Butyldimethylsilyloxy)-phenylmethyl]-15-hydroxymethyl-1,8,13,16-tetraoxadispiro[5.0.5.4]hexadecane 25 and ( $\left.6 R^{*}, 7 R^{*}, 14 R^{*}, 15 S^{*}\right)-14-\left[\left(S^{*}\right)\right.$ -(tert-butyldimethylsilyloxy)phenylmethyl]-15-hydroxymethyl-1,8,13,16-tetraoxadispiro[5.0.5.4]hexadecane 26
A solution of 9-borabicyclo[3.3.1]nonane (9-BBN) $(0.75 \mathrm{ml}$ of 0.5 M in THF, $0.375 \mathrm{mmol}, 4.9 \mathrm{eq}$.$) was added to 23(35.4 \mathrm{mg}$, 0.077 mmol ) and the resulting mixture was stirred at room temperature for 3 h . Then, the solution was cooled to $0^{\circ} \mathrm{C}$ and treated with $10 \% \mathrm{NaOH}$ solution $(0.4 \mathrm{ml})$ and $\mathrm{H}_{2} \mathrm{O}_{2}(30 \%, 0.4$ $\mathrm{ml})$ for 30 min . The reaction mixture was quenched with saturated $\mathrm{Na}_{2} \mathrm{CO}_{3}$ aqueous solution and the mixture was extracted with ether $(\times 3)$. The combined organic extracts were dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated in vacuo. The residue was purified by flash chromatography (eluent: ether-petrol $1: 3 \rightarrow 1: 1$; gradient) to give in order of elution $25(31.6 \mathrm{mg}, 0.066 \mathrm{mmol}, 86 \%)$ then 26 ( $5.3 \mathrm{mg}, 0.011 \mathrm{mmol}, 14 \%$ ).

25: $v_{\max }($ film $) / \mathrm{cm}^{-1} 3469,2949,2857,1465,1360,1254,1209$, $1162,1091,1035,854,838,770,733 ; \delta_{\mathrm{H}}\left(600 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.43$ ( $2 \mathrm{H}, \mathrm{d}, J 7.5, \mathrm{Ar}), 7.29(2 \mathrm{H}, \mathrm{t}, J 7.5, \mathrm{Ar}), 7.23(1 \mathrm{H}, \mathrm{t}, J 7.3, \mathrm{Ar})$, $4.97(1 \mathrm{H}, \mathrm{d}, J 9.2, \mathrm{PhCH}), 4.19(1 \mathrm{H}, \mathrm{d}, J 10.5, \mathrm{OH}), 4.14(1 \mathrm{H}$, dd, $J 9.2,4.1,14-\mathrm{H}), 4.08(1 \mathrm{H}, \mathrm{td}, J 11.9,3.6,15-\mathrm{H}), 4.05-3.99$ $\left(3 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OH}, 2_{\mathrm{ax}}-\mathrm{H}\right), 3.77-3.75\left(1 \mathrm{H}, \mathrm{m}, 2_{\mathrm{eq}}-\mathrm{H}\right), 3.22-3.19$ $\left(1 \mathrm{H}, \mathrm{m}, 9_{\mathrm{eq}}-\mathrm{H}\right), 2.37\left(1 \mathrm{H}, \mathrm{td}, J 10.6,1.9,9_{\mathrm{ax}}-\mathrm{H}\right), 1.85-1.15(12 \mathrm{H}$, $\mathrm{m}), 0.87\left(9 \mathrm{H}, \mathrm{s}, \mathrm{Bu}^{t}\right), 0.10(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}),-0.36(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}) ; \delta_{\mathrm{C}}(50$ $\mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) 142.6, 127.6, 127.5, 127.5 (Ar), 96.6, 94.6 (6-C, $7-\mathrm{C}), 74.2(\mathrm{PhCH}), 72.4,71.7$ (14-C, 15-C) 63.2, 61.8, 60.1 $\left(\mathrm{CH}_{2} \mathrm{OH}, 2-\mathrm{C}, 9-\mathrm{C}\right), 29.5,28.6,25.0,24.5,17.9,17.6$ (3-C, 4-C, 5-C, 10-C, 11-C, 12-C), $25.7\left(\mathrm{Bu}^{t}\right), 18.1\left(\mathrm{Bu}^{t}\right),-3.9(\mathrm{Me}),-5.5$ (Me); $m / z(\mathrm{CI}) 496\left(\mathrm{MNH}_{4}^{+}, 2 \%\right), 479\left(\mathrm{MH}^{+}, 11\right), 167$ (100), 118 (19) [Found $\left(\mathrm{MH}^{+}\right) 479.2829 . \mathrm{C}_{26} \mathrm{H}_{43} \mathrm{O}_{6} \mathrm{Si}$ requires $M \mathrm{H}$, 479.2829].

26: $v_{\max }($ film $) / \mathrm{cm}^{-1} 3474,2927,2856,1454,1360,1253,1210$, $1191,1161,1072,999,837,776,733,700 ; \delta_{\mathrm{H}}\left(600 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $7.39(2 \mathrm{H}, \mathrm{d}, J 7.4, \mathrm{Ar}), 7.29(2 \mathrm{H}, \mathrm{t}, J 7.4, \mathrm{Ar}), 7.22(1 \mathrm{H}, \mathrm{t}, J 7.3$, Ar), $4.77(1 \mathrm{H}, \mathrm{d}, J 6.0, \mathrm{PhCH}), 4.01-3.97(1 \mathrm{H}, \mathrm{m}, 15-\mathrm{H}), 3.81$ $(1 \mathrm{H}, \mathrm{dd}, J 9.4,6.1,14-\mathrm{H}), 3.77-3.73\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OH}\right), 3.73-$ $3.66(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H} \times 2), 3.54\left(1 \mathrm{H}, \mathrm{dd}, J 11.7,7.7, \mathrm{CH}_{2} \mathrm{OH}\right), 3.47-$ $3.43\left(1 \mathrm{H}, \mathrm{m}, 9_{\mathrm{eq}}-\mathrm{H}\right), 3.02-2.97\left(1 \mathrm{H}, \mathrm{m}, 9_{\mathrm{ax}}-\mathrm{H}\right), 2.17(1 \mathrm{H}, \mathrm{br}$, $\mathrm{OH}), 1.78-1.16(12 \mathrm{H}, \mathrm{m}), 0.90\left(9 \mathrm{H}, \mathrm{s}, \mathrm{Bu}^{t}\right), 0.08(3 \mathrm{H}, \mathrm{s}, \mathrm{Me})$, $-0.17(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}) ; \delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 142.4,127.6,127.2$, 126.9 (Ar), 96.0, 95.5 (6-C, 7-C), 75.7 (PhCH), 72.7, 69.9 (14-C, $15-\mathrm{C}), 62.7\left(\mathrm{CH}_{2} \mathrm{OH}\right), 60.6,60.5(2-\mathrm{C}, 9-\mathrm{C}), 29.6,28.2,24.8$, 24.8, 18.3, 17.7 (3-C, 4-C, 5-C, 10-C, 11-C, 12-C), 25.9 ( $\mathrm{Bu}^{t}$ ), $18.1\left(\mathrm{Bu}^{t}\right),-4.6(\mathrm{Me}),-4.8(\mathrm{Me}) ; m / z(\mathrm{CI}) 496\left(\mathrm{MNH}_{4}{ }^{+}, 17 \%\right)$, 287 (10), 167 (100), 118 (23) [Found $\left(\mathrm{MNH}_{4}^{+}\right) 496.3094$. $\mathrm{C}_{26} \mathrm{H}_{46} \mathrm{O}_{6} \mathrm{NSi}$ requires $\left.M \mathrm{NH}_{4}, 496.3094\right]$.
(2R,6S,7R,9R,15S)-15-[(S)-(tert-Butyldimethylsilyloxy)phenyl-methyl]-2,9-bis(phenylthiomethyl)-1,8,13,16-tetraoxadispiro-[5.0.5.4]hexadecan-14-one 22

A solution of $7 \mathbf{a}(44.7 \mathrm{mg}, 0.075 \mathrm{mmol})$ in DMF ( 1 ml ) containing imidazole ( $24.2 \mathrm{mg}, 0.36 \mathrm{mmol}, 4.8$ eq.) and tert-butyldimethylsilyl chloride ( $36.1 \mathrm{mg}, 0.24 \mathrm{mmol}, 3.2 \mathrm{eq}$.$) was stirred$ at room temperature for 7 h . To complete the reaction, more imidazole ( $39.5 \mathrm{mg}, 0.58 \mathrm{mmol}, 7.7$ eq.) and tert-butyldimethylsilyl chloride ( $54.1 \mathrm{mg}, 0.36 \mathrm{mmol}, 4.8 \mathrm{eq}$. ) were added to the solution and the mixture was stirred for another 17 h . The reaction was quenched with $\mathrm{H}_{2} \mathrm{O}$ and the mixture was extracted with ether $(\times 3)$. The combined organic extracts were
dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated in vacuo. Flash chromatography (eluent: ether-petrol $1: 5 \rightarrow 1: 3$; gradient) gave the title compound 22 as a colourless oil ( $35.8 \mathrm{mg}, 0.051 \mathrm{mmol}, 67 \%$ ); $[\alpha]_{\mathrm{D}}^{23}$ $-31.0\left(c 1.08, \mathrm{CHCl}_{3}\right) ; v_{\max }($ film $) / \mathrm{cm}^{-1} 2951,2855,2360,1744$, $1585,1481,1439,1257,1205,1104,1051,992,957,910,837$, $778,736,701 ; \delta_{\mathrm{H}}\left(600 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.39-7.10(15 \mathrm{H}, \mathrm{m}, \mathrm{Ar})$, $5.32(1 \mathrm{H}, \mathrm{d}, J 4.0, \mathrm{PhCH}), 4.88(1 \mathrm{H}, \mathrm{d}, J 4.0,15-\mathrm{H}), 3.82-3.78$ $(1 \mathrm{H}, \mathrm{m}, 9-\mathrm{H}), 3.69(1 \mathrm{H}, \mathrm{dtd}, J 11.7,6.0,2.3,2-\mathrm{H}), 3.04(1 \mathrm{H}, \mathrm{dd}$, $\left.J 13.8,7.1, \mathrm{PhSCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}\right), 2.96\left(1 \mathrm{H}, \mathrm{dd}, J 13.8,5.0, \mathrm{PhSCH}_{\mathrm{A}} H_{\mathrm{B}}\right)$, $2.52\left(2 \mathrm{H}, \mathrm{d}, J 5.6, \mathrm{CH}_{2} \mathrm{SPh}\right), 1.83-1.54(8 \mathrm{H}, \mathrm{m}), 1.37-1.19(4 \mathrm{H}$, $\mathrm{m}), 0.87\left(9 \mathrm{H}, \mathrm{s}, \mathrm{Bu}^{t}\right), 0.05(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}),-0.14(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}) ; \delta_{\mathrm{C}}(50$ $\left.\mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 166.8(\mathrm{C}=\mathrm{O}), 140.1,136.9,136.8,128.8,128.7$, $128.0,127.3,125.8,125.7$ (Ar), 103.8, 96.0 (6-C, 7-C), 74.2, 73.9 $(15-\mathrm{C}, \mathrm{PhCH}), 71.9,70.6$ (2-C, 9-C), $39.1\left(\mathrm{CH}_{2} \mathrm{SPh}\right), 38.4$ $\left(\mathrm{CH}_{2} \mathrm{SPh}\right), 29.8,28.4,27.9,27.3,18.1,16.9$ (3-C, 4-C, 5-C, 10-C, $11-\mathrm{C}, 12-\mathrm{C}), 25.6\left(\mathrm{Bu}^{t}\right), 18.0\left(\mathrm{Bu}^{t}\right),-5.0(\mathrm{Me}),-5.1(\mathrm{Me}) ;$ $m / z(\mathrm{CI}) 707\left(\mathrm{MH}^{+}, 7 \%\right), 411$ (100), 303 (15) [Found ( $\mathrm{MH}^{+}$) 707.2900. $\mathrm{C}_{39} \mathrm{H}_{51} \mathrm{O}_{6} \mathrm{SiS}_{2}$ requires $M \mathrm{H}, 707.2896$ ].
(2R,6R,7S,9R,14S)-14-[(S)-(tert-Butyldimethylsilyloxy)phenyl-methyl]-15-methylene-2,9-bis(phenylthiomethyl)-1,8,13,16-tetraoxadispiro[5.0.5.4]hexadecane 24

To a solution of $22(34.7 \mathrm{mg}, 0.049 \mathrm{mmol})$ in THF $(2.5 \mathrm{ml})$ containing pyridine $(0.5 \mathrm{ml})$ was added 0.5 M solution of the Tebbe reagent in toluene ( $0.15 \mathrm{ml}, 0.075 \mathrm{mmol}, 1.5 \mathrm{eq}$.$) at$ $-78^{\circ} \mathrm{C}$. After stirring for 5 min , the mixture was warmed up to room temperature and stirred for 30 min . Then, the mixture was quenched with a saturated $\mathrm{Na}_{2} \mathrm{CO}_{3}$ solution $(0.5 \mathrm{ml})$ and ether at $0^{\circ} \mathrm{C}$. The formed solid was filtered over Celite and washed with ether. The organic filtrate was dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated in vacuo. The resulting residue was purified by flash chromatography (eluent: ether-petrol $1: 7 \rightarrow 1: 5$; gradient) to give the title compound 24 ( $30.3 \mathrm{mg}, 0.043 \mathrm{mmol}, 88 \%$ ); $[a]_{\mathrm{D}}^{23}-11.1(c$ $\left.1.57, \mathrm{CHCl}_{3}\right) ; v_{\text {max }}(\mathrm{film}) / \mathrm{cm}^{-1} 2949,2928,2360,1662,1585$, $1481,1439,1257,1201,1167,1091,1040,910,855,838,777$, 736,$700 ; \delta_{\mathrm{H}}\left(600 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.37-7.14(15 \mathrm{H}, \mathrm{m}, \mathrm{Ar}), 4.84$ $(1 \mathrm{H}, \mathrm{d}, J 6.8, \mathrm{PhCH}), 4.64\left(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2}=\right), 4.46\left(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2}=\right)$, $4.32(1 \mathrm{H}, \mathrm{d}, J 6.8,14-\mathrm{H}), 4.10-4.05(1 \mathrm{H}, \mathrm{m}, 9-\mathrm{H}$ or $2-\mathrm{H}), 3.40-$ $3.35(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}$ or $9-\mathrm{H}), 3.12\left(1 \mathrm{H}, \mathrm{dd}, J 13.5,4.6, \mathrm{PhSCH}_{\mathrm{A}^{-}}\right.$ $\left.\mathrm{H}_{\mathrm{B}}\right), 3.05\left(1 \mathrm{H}\right.$, dd, $\left.J 13.5,6.3, \mathrm{PhSCH}^{\prime}{ }_{\mathrm{A}} \mathrm{H}^{\prime}{ }_{\mathrm{B}}\right), 2.86(1 \mathrm{H}$, dd, $J$ 13.5, 5.9, $\left.\mathrm{PhSCH}_{\mathrm{A}}^{\prime} H^{\prime}{ }_{\mathrm{B}}\right), 2.84(1 \mathrm{H}, \mathrm{dd}, J 13.5,7.7, \mathrm{PhS}-$ $\left.\mathrm{CH}_{\mathrm{A}} H_{\mathrm{B}}\right), 1.90-1.03(12 \mathrm{H}, \mathrm{m}), 0.90\left(9 \mathrm{H}, \mathrm{s}, \mathrm{Bu}^{t}\right), 0.11(3 \mathrm{H}, \mathrm{s}$, $\mathrm{Me}),-0.28(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}) ; \delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 152.5$ (15-C), $142.1,137.3,137.0,128.8,128.7,128.5,128.5,127.5,127.4$, $127.0,125.5,125.5(\mathrm{Ar}), 99.5,97.2(6-\mathrm{C}, 7-\mathrm{C}), 95.4\left(\mathrm{CH}_{2}=\right)$, 74.8, 70.7, 69.7, 69.0 (2-C, 9-C, 14-C, PhCH$), 39.0\left(\mathrm{CH}_{2} \mathrm{SPh}\right)$, $38.7\left(\mathrm{CH}_{2} \mathrm{SPh}\right), 29.8,29.4,27.8,27.7,17.9,17.8$ (3-C, 4-C, 5-C, $10-\mathrm{C}, 11-\mathrm{C}, 12-\mathrm{C}), 25.8\left(\mathrm{Bu}^{t}\right), 18.1\left(\mathrm{Bu}^{t}\right),-4.3(\mathrm{Me}),-4.8(\mathrm{Me}) ;$ $m / z(\mathrm{CI}) 705\left(\mathrm{MH}^{+}, 1 \%\right), 573$ (6), 411 (17) , 221 (100) [Found $\left(\mathrm{MH}^{+}\right) 705.3100 . \mathrm{C}_{40} \mathrm{H}_{53} \mathrm{O}_{5} \mathrm{SiS}_{2}$ requires $\left.M \mathrm{H}, 705.3103\right]$.
(2R,6S,7S,9R,14R,15R)-14-[(S)-(tert-Butyldimethylsilyloxy)-phenylmethyl]-15-hydroxymethyl-2,9-bis(phenylthiomethyl)-$1,8,13,16$-tetraoxadispiro[5.0.5.4]hexadecane 27 and ( $2 R, 6 S$, 7S,9R,14R,15S)-14-[(S)-(tert-butyldimethylsilyloxy)phenyl-methyl]-15-hydroxymethyl-2,9-bis(phenylthiomethyl)-1,8,13,16tetraoxadispiro[5.0.5.4]hexadecane 28
A solution of $9-\mathrm{BBN}(1.0 \mathrm{ml}$ of 0.5 M in THF, $0.50 \mathrm{mmol}, 5.7$ eq.) was added to $24(61.6 \mathrm{mg}, 0.0873 \mathrm{mmol})$, and stirred at room temperature for 3 h . Then, the resulting solution was cooled to $0^{\circ} \mathrm{C}$ and treated with a $10 \% \mathrm{NaOH}$ solution $(0.5 \mathrm{ml})$ and $\mathrm{H}_{2} \mathrm{O}_{2}(30 \%, 0.5 \mathrm{ml})$ for 30 min . The reaction mixture was quenched with a saturated $\mathrm{Na}_{2} \mathrm{CO}_{3}$ aqueous solution and the mixture was extracted with ether $(\times 3)$. The combined organic extracts were dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated in vacuo. The residue was purified by flash chromatography (eluent: ether-petrol $1: 5 \rightarrow 1: 2$; gradient) to give in order of elution $27(37.9 \mathrm{mg}$, $0.0524 \mathrm{mmol}, 60 \%)$ then $28(12.8 \mathrm{mg}, 0.0177 \mathrm{mmol}, 20 \%)$.

27: $[a]_{\mathrm{D}}^{23}-0.2\left(c 1.00, \mathrm{CHCl}_{3}\right) ; v_{\max }($ film $) / \mathrm{cm}^{-1} 3472,2928$,

2855, 1584, 1480, 1438, 1254, 1191, 1090, 1011, 982, 910, 856, $838,777,736,690 ; \delta_{\mathrm{H}}\left(600 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.40-7.14(15 \mathrm{H}, \mathrm{m}$, Ar), $4.89(1 \mathrm{H}, \mathrm{d}, J 8.8, \mathrm{PhCH}), 4.24(1 \mathrm{H}, \mathrm{dd}, J 8.8,4.1,14-\mathrm{H})$, 4.16-4.12 ( $1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}), 4.06-4.02\left(1 \mathrm{H}, \mathrm{m}, \mathrm{C} H_{\mathrm{A}} \mathrm{H}_{\mathrm{B}} \mathrm{OH}\right), 3.92-$ $3.89\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{\mathrm{A}} H_{\mathrm{B}} \mathrm{OH}, 15-\mathrm{H}\right), 3.82(1 \mathrm{H}, \mathrm{d}, J 10.3, \mathrm{OH}), 3.22$ $\left(1 \mathrm{H}, \mathrm{dd}, J 13.3,3.9, \mathrm{PhSC}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}\right), 2.94(1 \mathrm{H}, \mathrm{dd}, J 13.3,7.9$, $\mathrm{PhSCH}_{\mathrm{A}} H_{\mathrm{B}}$ ), $2.88\left(1 \mathrm{H}, \mathrm{dd}, J 13.2,6.8, \mathrm{PhSCH}^{\prime}{ }_{\mathrm{A}} \mathrm{H}^{\prime}{ }_{\mathrm{B}}\right.$ ), 2.75 $\left(1 \mathrm{H}, \mathrm{dd}, J 13.2,5.1, \mathrm{PhSCH}^{\prime}{ }_{\mathrm{A}} H^{\prime}{ }_{\mathrm{B}}\right), 2.66-2.62(1 \mathrm{H}, \mathrm{m}, 9-\mathrm{H})$, $1.90-1.00(12 \mathrm{H}, \mathrm{m}), 0.88\left(9 \mathrm{H}, \mathrm{s}, \mathrm{Bu}^{\prime}\right), 0.09(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}),-0.32$ $(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}) ; \delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 142.1,137.7,136.4,128.8$, 128.7, 128.7, 127.6, 127.6, 127.5, 125.8, 125.5 (Ar), 97.6, 95.8 (6-C, 7-C), 74.5, 72.0, 71.6, 70.1, 69.1 (2-C, 9-C, 14-C, 15-C, $\mathrm{PhCH}), 63.0\left(\mathrm{CH}_{2} \mathrm{OH}\right), 39.3\left(\mathrm{CH}_{2} \mathrm{SPh}\right), 38.7\left(\mathrm{CH}_{2} \mathrm{SPh}\right)$, 29.5, 29.4, 29.2, 28.1, 17.9, 17.8 (3-C, 4-C, 5-C, 10-C, 11-C, $12-\mathrm{C}), 25.7\left(\mathrm{Bu}^{t}\right), 17.9\left(\mathrm{Bu}^{t}\right),-4.0(\mathrm{Me}),-5.4(\mathrm{Me}) ; m / z(\mathrm{CI})$ $740\left(\mathrm{MNH}_{4}^{+}, 10 \%\right), 723\left(\mathrm{MH}^{+}, 16\right), 705(32), 411$ (100), 303 (47) [Found $\left(\mathrm{MH}^{+}\right) 723.3210 . \mathrm{C}_{40} \mathrm{H}_{55} \mathrm{O}_{6} \mathrm{SiS}_{2}$ requires $M \mathrm{H}$, 723.3209].

28: $[a]_{\mathrm{D}}^{23}-16.4\left(c 1.22, \mathrm{CDCl}_{3}\right) ; v_{\max }($ film $) / \mathrm{cm}^{-1} 3424,2927$, $2855,1584,1480,1438,1258,1200,1104,1036,836,777,737$, $690 ; \delta_{\mathrm{H}}\left(600 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.34-7.14(15 \mathrm{H}, \mathrm{m}, \mathrm{Ar}), 4.67-4.65$ $(1 \mathrm{H}, \mathrm{m}, \mathrm{PhCH}), 3.94-3.91(2 \mathrm{H}, \mathrm{m}, 14-\mathrm{H}, 15-\mathrm{H}), 3.84-3.80(1 \mathrm{H}$, $\mathrm{m}, 2-\mathrm{H}), 3.60-3.55\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}} \mathrm{OH}\right), 3.48-3.42(1 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{\mathrm{A}} H_{\mathrm{B}} \mathrm{OH}\right), 3.37-3.32(1 \mathrm{H}, \mathrm{m}, 9-\mathrm{H}), 3.06(1 \mathrm{H}, \mathrm{dd}, J 13.4,6.0$, $\mathrm{PhSCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}$ ), $3.01\left(1 \mathrm{H}, \mathrm{dd}, J 13.5,7.2, \mathrm{PhSCH}^{\prime}{ }_{\mathrm{A}} \mathrm{H}^{\prime}{ }_{\mathrm{B}}\right), 2.93$ $\left(1 \mathrm{H}, \mathrm{dd}, J 13.4,6.0, \mathrm{PhSCH}_{\mathrm{A}} H_{\mathrm{B}}\right), 2.85(1 \mathrm{H}, \mathrm{dd}, J 13.5,5.0$, $\left.\mathrm{PhSCH}_{\mathrm{A}}{ }^{\prime} H^{\prime}{ }_{\mathrm{B}}\right), 1.91(1 \mathrm{H}, \mathrm{dd}, J 7.9,5.1, \mathrm{OH}), 1.75-1.06(12 \mathrm{H}$, m), $0.90\left(9 \mathrm{H}, \mathrm{s}, \mathrm{Bu}^{t}\right), 0.08(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}),-0.16(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}) ; \delta_{\mathrm{C}}(50$ $\mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) $142.0,137.6,137.3,129.0,128.7,128.7,128.4$, 127.6, 127.1, 127.0, 125.7, 125.4 (Ar), 96.6, 96.3 (6-C, 7-C), 75.9, 72.0, 69.8, 69.7, 69.3 (2-C, 9-C, 14-C, 15-C, PhCH), 62.6 $\left(\mathrm{CH}_{2} \mathrm{OH}\right), 39.3\left(\mathrm{CH}_{2} \mathrm{SPh}\right), 39.1\left(\mathrm{CH}_{2} \mathrm{SPh}\right), 30.1,29.6,27.6$, 27.6, 18.3, 18.0 (3-C, 4-C, 5-C, 10-C, 11-C, 12-C), 25.9 ( $\mathrm{Bu}^{t}$ ), $18.2\left(\mathrm{Bu}^{+}\right),-4.7(\mathrm{Me}) ; m / z(\mathrm{CI}) 740\left(\mathrm{MNH}_{4}{ }^{+}, 12 \%\right), 411(11)$, 132 (53), 126 (56), 52 (100) [Found ( $\mathrm{MNH}_{4}{ }^{+}$) 740.3480. $\mathrm{C}_{40} \mathrm{H}_{58} \mathrm{O}_{6} \mathrm{NSiS}_{2}$ requires $\left.M \mathrm{NH}_{4}, 740.3475\right]$.

## (2R,6S,7S,9R,14R,15R)-14-[(S)-(tert-Butyldimethylsilyloxy)-phenylmethyl]-15-hydroxymethyl-2,9-bis(phenylsulfonylmethyl)-1,8,13,16-tetraoxadispiro[5.0.5.4]hexadecane 29

To a solution of $27(37.9 \mathrm{mg}, 0.0524 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \mathrm{ml})$, was added MCBPA ( $50-60 \%$ purity, $90 \mathrm{mg}, 0.26 \mathrm{mmol}$ ) at $0^{\circ} \mathrm{C}$. The mixture was allowed to warm up to room temperature and stirred for 1.5 h . The reaction was quenched with a saturated $\mathrm{Na}_{2} \mathrm{SO}_{3}$ aqueous solution at $0{ }^{\circ} \mathrm{C}$, and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(\times 4)$. The combined organic phases were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporated in vacuo. The resulting residue was purified by flash chromatography (eluent: neat ether) to afford the title compound $29(36.3 \mathrm{mg}, 0.0461 \mathrm{mmol}, 88 \%)$ as a white solid; mp $118^{\circ} \mathrm{C}$ (ether); $[a]_{\mathrm{D}}^{23}-41.7\left(c 1.54, \mathrm{CHCl}_{3}\right) ; v_{\max }($ film $) / \mathrm{cm}^{-1} 3520$, 3056, 2933, 1770, 1586, 1447, 1365, 1303, 1261, 1202, 1148 , 1087, 1067, 1039, 1006, 970, 909, 860, 838, 780, 735, 688; $\delta_{\mathrm{H}}\left(600 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.93(2 \mathrm{H}, \mathrm{d}, J 7.6, \mathrm{Ar}), 7.89(2 \mathrm{H}, \mathrm{d}, J 7.6$, $\mathrm{Ar}), 7.67(1 \mathrm{H}, \mathrm{t}, J 7.4, \mathrm{Ar}), 7.60-7.53(5 \mathrm{H}, \mathrm{m}, \mathrm{Ar}), 7.49(2 \mathrm{H}, \mathrm{t}$, $J 7.7, \mathrm{Ar}), 7.35(2 \mathrm{H}, \mathrm{t}, J 7.6, \mathrm{Ar}), 7.26(1 \mathrm{H}, \mathrm{t}, J 7.3, \mathrm{Ar}), 4.81$ $(1 \mathrm{H}, \mathrm{d}, J 8.5, \mathrm{PhC} H), 4.48(1 \mathrm{H}, \mathrm{td}, J 8.5,2.8,2-\mathrm{H}), 4.25(1 \mathrm{H}$, dd, $J 8.5,3.9,14-\mathrm{H}), 4.02-3.97\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}} \mathrm{OH}\right), 3.88-3.85$ $(1 \mathrm{H}, \mathrm{m}, 15-\mathrm{H}), 3.72\left(1 \mathrm{H}, \mathrm{dd}, J 12.1,2.0, \mathrm{CH}_{\mathrm{A}} H_{\mathrm{B}} \mathrm{OH}\right), 3.40(1 \mathrm{H}$, br, OH ), $3.33\left(1 \mathrm{H}, \mathrm{dd}, J 14.1,3.3, \mathrm{PhSO}_{2} \mathrm{CH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}\right.$ ), 3.29-3.22 $\left(2 \mathrm{H}, \mathrm{m}, \mathrm{PhSCH}^{\prime}{ }_{\mathrm{A}} \mathrm{H}^{\prime}{ }_{\mathrm{B}}, 9-\mathrm{H}\right), 3.15\left(1 \mathrm{H}, \mathrm{dd}, J 14.1,8.3, \mathrm{PhSCH}_{\mathrm{A}}{ }^{-}\right.$ $\left.H_{\mathrm{B}}\right), 3.00-2.95\left(1 \mathrm{H}, \mathrm{m}, \mathrm{PhSCH}_{\mathrm{A}} H^{\prime}{ }_{\mathrm{B}}\right), 1.84(1 \mathrm{H}$, br d, $J 12.7)$, $1.64-0.87(10 \mathrm{H}, \mathrm{m}), 0.92\left(9 \mathrm{H}, \mathrm{s}, \mathrm{Bu}^{\dagger}\right), 0.59(1 \mathrm{H}, \mathrm{br} \mathrm{d}, J 13.7)$, $0.08(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}),-0.25(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}) ; \delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 141.4$, 141.1, 139.8, 133.7, 132.9, 129.3, 128.9, 127.9, 127.8, 127.7, 127.7, 127.6 (Ar), 97.2, 95.6 (6-C, 7-C), 74.7, 72.0, 71.1, 65.8, 64.9 (2-C, $9-\mathrm{C}, 14-\mathrm{C}, 15-\mathrm{C}, \mathrm{PhCH}), 61.9,61.9,61.6\left(\mathrm{CH}_{2} \mathrm{OH}\right.$, $2 \times \mathrm{CH}_{2} \mathrm{SPh}$ ), $30.9,29.7,28.6,27.9,17.5,17.0$ (3-C, 4-C, 5-C, $10-\mathrm{C}, 11-\mathrm{C}, 12-\mathrm{C}), 25.8\left(\mathrm{Bu}^{+}\right) 18.0\left(\mathrm{Bu}^{t}\right),-4.1(\mathrm{Me}),-5.3(\mathrm{Me}) ;$ $\mathrm{m} / \mathrm{z}(\mathrm{CI}) 804\left(\mathrm{MNH}_{4}{ }^{+}, 45 \%\right), 787\left(\mathrm{MH}^{+}, 92\right), 672(58), 647(75)$,

492 (95), 475 (100) [Found $\left(\mathrm{MH}^{+}\right) 787.3010 . \mathrm{C}_{40} \mathrm{H}_{55} \mathrm{O}_{10} \mathrm{SiS}_{2}$ requires $M \mathrm{H}, 787.3006]$.

## ( $2 R, 3 R, 4 S$ )-4-(tert-Butyldimethylsilyloxy)-4-phenylbutane-1,2,3-triol 30

To a stirred solution of $29(35.3 \mathrm{mg}, 0.045 \mathrm{mmol})$ in THF ( 1.0 $\mathrm{ml})$ under argon was added a solution of lithium bis(trimethylsilyl)amide ( 1.0 M in THF, $0.12 \mathrm{ml}, 0.12 \mathrm{mmol}, 2.5 \mathrm{eq}$.) at $0^{\circ} \mathrm{C}$, and the resulting mixture was stirred for 30 min at room temperature. The reaction mixture was quenched with saturated $\mathrm{NH}_{4} \mathrm{Cl}$ aquous solution and then extracted with ether ( $\times 3$ ). The combined organic extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated in vacuo. The resulting residue was purified by flash chromatography (eluent: ether-petrol $1: 1 \rightarrow$ neat ether; gradient) to give the title compound $\mathbf{3 0}(7.5 \mathrm{mg}, 0.024 \mathrm{mmol}, 53 \%)$ as a colourless oil; $[a]_{\mathrm{D}}^{23}+32.0\left(c 0.75, \mathrm{CHCl}_{3}\right) ; v_{\max }($ film $) / \mathrm{cm}^{-1} 3415$, 2954, 2929, 2857, 1635, 1472, 1389, 1306, 1254, 1085, 910, 838, 779,$734 ; \delta_{\mathrm{H}}\left(600 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.39-7.29(5 \mathrm{H}, \mathrm{m}, \mathrm{Ar}), 4.79$ ( $1 \mathrm{H}, \mathrm{d}, J 6.2,4-\mathrm{H}$ ), $3.84-3.78(3 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{H}-1,3-\mathrm{H}), 3.60(1 \mathrm{H}$, td, $J 6.7,4.2,2-\mathrm{H}), 2.99(1 \mathrm{H}, \mathrm{br}, \mathrm{OH}), 2.27(1 \mathrm{H}, \mathrm{br}, \mathrm{OH}), 1.60$ $(1 \mathrm{H}, \mathrm{br}, \mathrm{OH}), 0.89\left(9 \mathrm{H}, \mathrm{s}, \mathrm{Bu}^{\prime}\right), 0.06(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}),-0.19(3 \mathrm{H}, \mathrm{s}$, $\mathrm{Me}) ; \delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 140.4,128.3,128.1,127.3$ (Ar), 77.2, 76.4, 71.9 (2-C, 3-C, 4-C), 63.9 (1-C), $25.7\left(\mathrm{Bu}^{t}\right), 18.0\left(\mathrm{Bu}^{t}\right)$, -4.6 (Me), -5.2 (Me); $m / z$ (CI) $330\left(\mathrm{MNH}_{4}^{+}, 10 \%\right), 313$ $\left(\mathrm{MH}^{+}, 19 \%\right), 198$ (100) [Found ( $\mathrm{MH}^{+}$) 313.1835. $\mathrm{C}_{16} \mathrm{H}_{29} \mathrm{O}_{4} \mathrm{Si}$ requires $M \mathrm{H}, 313.1835]$.

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## References and notes

1 Dispiroketals in synthesis. Part 24: D. Lainé, M. Fujita and S. V. Ley, J. Chem. Soc., Perkin Trans. 1, 1999, 1639.
2 P. M. Collins and R. J. Ferrier, Monosaccharides, John Wiley \& Sons, Chichester, 1995
3 Y. Hirata, D. Uemura, K. Ueda and S. Takano, Pure Appl. Chem., 1979, 51, 1875; D. Uemura, K. Ueda, Y. Hirata, H. Naoki and T. Iwashita, Tetrahedron Lett., 1981, 22, 1909, 2781; P. J. Scheuer and R. E. Moore, Science, 1971, 172, 495; R. E. Moore and G. Bartolini, J. Am. Chem. Soc., 1981, 103, 2491.

4 N. Minami, S. S. Ko and Y. Kishi, J. Am. Chem. Soc., 1982, 104, 1109; A. W. M. Lee, V. S. Martin, S. Masamune, K. B. Sharpless and F. J. Walker, J. Am. Chem. Soc., 1982, 104, 3315; S. Y. Ko, A. W. M. Lee, S. Masamune, L. A. Reed, III, K. B. Sharpless and F. J. Walker, Science, 1983, 220, 949.
5 For a comprehensive review of dispiroketal chemistry see: S. V. Ley, R. Downham, P. J. Edwards, J. E. Innes and M. Woods, Contemp. Org. Synth., 1995, 2, 365.
6 (a) S. V. Ley, S. Mio and B. Meseguer, Synlett, 1996, 787; (b) S. V. Ley and S. Mio, Synlett, 1996, 789; (c) S. V. Ley, S. Mio and B. Meseguer, Synlett, 1996, 791.

7 N. L. Douglas, S. V. Ley, U. Lücking and S. L. Warriner, J. Chem. Soc., Perkin Trans. 1, 1998, 51; G.-J. Boons, P. Grice, R. Leslie, S. V. Ley and L. L. Yeung, Tetrahedron Lett., 1993, 34, 8523.

8 B. C. B. Bezuidenhoudt, G. H. Castle and S. V. Ley, Tetrahedron Lett., 1994, 35, 7447; B. C. B. Bezuidenhoudt, G. H. Castle, J. V. Geden and S. V. Ley, Tetrahedron Lett., 1994, 35, 7451; G. H. Castle and S. V. Ley, Tetrahedron Lett., 1994, 35, 7455.
9 In referring to diastereoisomers obtained from reaction of nonchiral starting materials, their racemic nature is assumed and only one enantiomeric form is depicted in their formulae.
10 G.-J. Boons, R. Downham, K. S. Kim, S. V. Ley and M. Woods, Tetrahedron, 1994, 50, 7157.
11 R. P. Linstead, L. N. Owen and R. F. Webb, J. Chem. Soc., 1953, 1218.

12 J.-N. Denis, A. Correa and A. E. Greene, J. Org. Chem., 1990, 55, 1957; Z.-M. Wang, H. C. Kolb and K. B. Sharpless, J. Org. Chem., 1994, 59, 5104.

13 B. R. Matthews, W. R. Jackson, H. A. Jacobs and K. G. Watson, Aust. J. Chem., 1990, 43, 1195.
14 During further studies of these aldol reactions, it was found that replacing DMPU by HMPA enhanced the nucleophilicity of the enolate of $\mathbf{1 8}$ (antipode of 4). Under these conditions it was possible


Reagents and conditions: $\mathrm{i}, \operatorname{Pr}^{\mathrm{i}} \mathrm{NH}, \mathrm{Bu} \mathrm{Li}, \mathrm{THF}-\mathrm{HMPA}$ ( $3: 1$ mixture), $-78^{\circ} \mathrm{C}, 30 \mathrm{~min}$ then pivaldehyde at $-78^{\circ} \mathrm{C}, 30 \mathrm{~min}$.
to carry out the reaction with pivalaldehyde and a single compound was isolated in $58 \%$ yield. The stereochemistry of this compound was not formally established but was predicted to be erythro according to the transition states shown in Scheme 2.
15 For a general review, see: J. Jurczak, S. Pikul and T. Bauer, Tetrahedron, 1986, 42, 447.
16 For a general review, see: S. Masamune, W. Choy, J. S. Petersen and L. R. Site, Angew. Chem., Int. Ed. Engl., 1985, 24, 1.

17 (a) C. H. Heathcock, C. T. White, J. J. Morrison and D. vanDerveer, J. Org. Chem., 1981, 46, 1296; (b) C. H. Heathcock and C. T. White, J. Am. Chem. Soc., 1979, 101, 7076; (c) C. H. Heathcock, M. C. Pirrung, C. T. Buse, J. P. Hagan, S. D. Young and J. E. Sohn, J. Am. Chem. Soc., 1979, 101, 7077.
18 M. Chérest, H. Felkin and N. Prudent, Tetrahedron Lett., 1968, 2199; N. T. Anh, Top. Curr. Chem., 1980, 88, 145.
19 C. H. Heathcock, S. D. Young, J. P. Hagan, M. C. Pirrung, C. T. White and D. vanDerveer, J. Org. Chem., 1980, 45, 3846.
20 M. T. Reetz, Angew. Chem., Int. Ed. Engl., 1984, 23, 556.
21 Y. Senda, S. Kamiyama and S. Imaizumi, Tetrahedron, 1977, 33, 2933.

22 E. Baer and H. O. L. Fischer, J. Biol. Chem., 1939, 128, 463; S. Hagen, T. Anthonsen and L. Kilaas, Tetrahedron, 1979, 35, 2583.


[^0]:    $\dagger$ CCDC reference number 207/326. See http://www.rsc.org/suppdata/ pl/1999/1647 for crystallographic files in .cif format.

